

Compressible Flow - TME085

Lecture 3

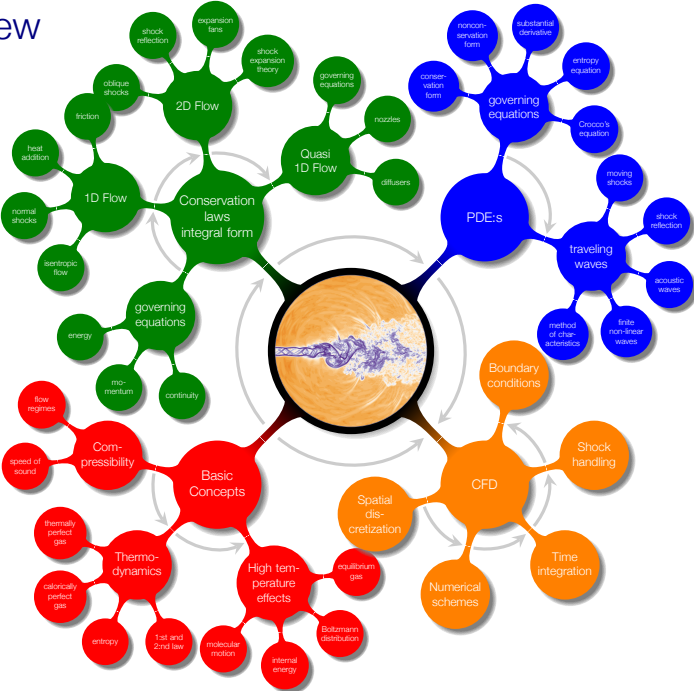
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Overview

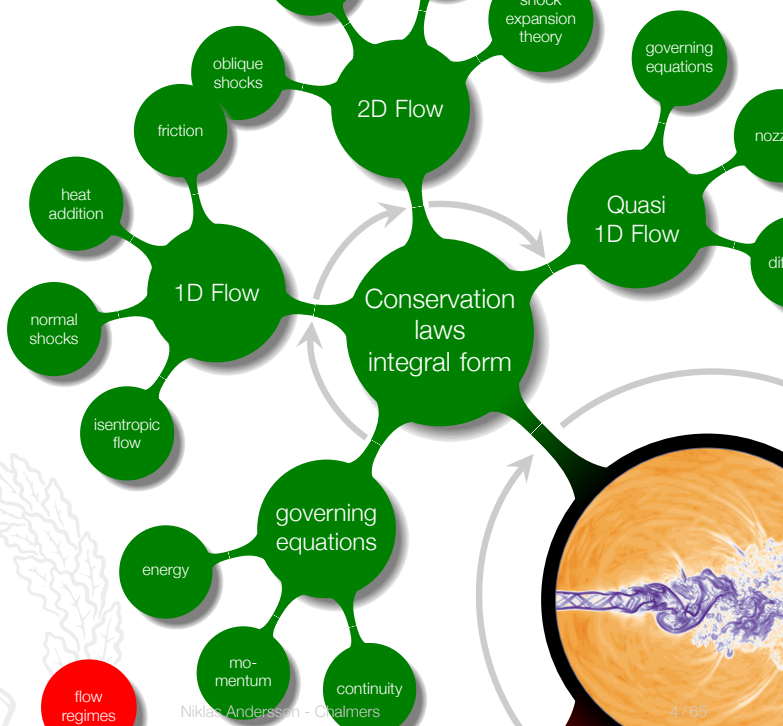


Chapter 3

One-Dimensional Flow



Overview

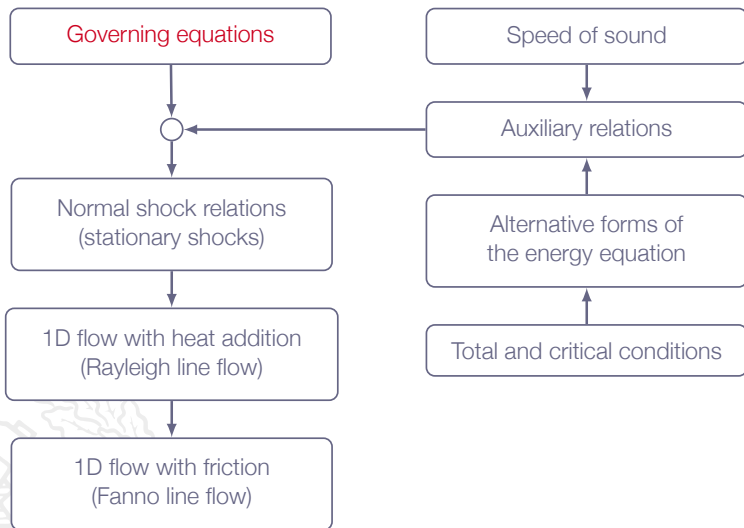


Addressed Learning Outcomes

- 4 **Present** at least two different formulations of the governing equations for compressible flows and **explain** what basic conservation principles they are based on
- 5 **Explain** how thermodynamic relations enter into the flow equations
- 6 **Define** the special cases of calorically perfect gas, thermally perfect gas and real gas and **explain** the implication of each of these special cases
- 8 **Derive** (marked) and **apply** (all) of the presented mathematical formulae for classical gas dynamics
 - a 1D isentropic flow*
 - b normal shocks*

one-dimensional flows - isentropic and non-isentropic

Roadmap - One-dimensional Flow

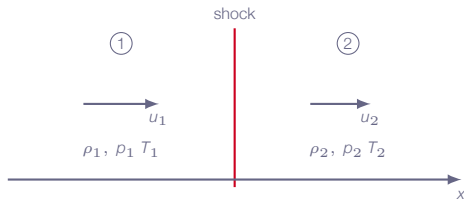


Chapter 3.2

One-Dimensional Flow Equations



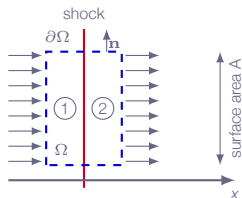
One-Dimensional Flow Equations



Assumptions:

- ▶ all flow variables only depend on x
- ▶ velocity aligned with x -axis

One-Dimensional Flow Equations



Control volume approach:

Define a rectangular control volume around shock, with upstream conditions denoted by 1 and downstream conditions by 2

One-Dimensional Flow Equations

Conservation of mass:

$$\frac{d}{dt} \iiint_{\Omega} \rho d\mathcal{V} = 0$$

$$\oiint_{\partial\Omega} \rho \mathbf{v} \cdot \mathbf{n} dS = \rho_2 u_2 A - \rho_1 u_1 A$$

$$\rho_1 u_1 = \rho_2 u_2$$

Conservation of momentum:

$$\frac{d}{dt} \iiint_{\Omega} \rho \mathbf{v} d\mathcal{V} = 0$$

$$\oiint_{\partial\Omega} [\rho(\mathbf{v} \cdot \mathbf{n})\mathbf{v} + p\mathbf{n}] dS =$$
$$(\rho_2 u_2^2 + p_2)A - (\rho_1 u_1^2 + p_1)A$$

$$\rho_1 u_1^2 + p_1 = \rho_2 u_2^2 + p_2$$

One-Dimensional Flow Equations

Conservation of energy:

$$\frac{d}{dt} \iiint_{\Omega} \rho e_o d\mathcal{V} = 0$$

$$\begin{aligned} \iint_{\partial\Omega} [\rho h_o \mathbf{v} \cdot \mathbf{n}] dS = \\ \rho_2 h_{o_2} u_2 A - \rho_1 h_{o_1} u_1 A \end{aligned}$$

$$\rho_1 u_1 h_{o_1} = \rho_2 u_2 h_{o_2}$$

Using the continuity equation this reduces to

$$h_{o_1} = h_{o_2}$$

or

$$h_1 + \frac{1}{2} u_1^2 = h_2 + \frac{1}{2} u_2^2$$

One-Dimensional Flow Equations

Summary:

$$\rho_1 u_1 = \rho_2 u_2$$

$$\rho_1 u_1^2 + p_1 = \rho_2 u_2^2 + p_2$$

$$h_1 + \frac{1}{2}u_1^2 = h_2 + \frac{1}{2}u_2^2$$

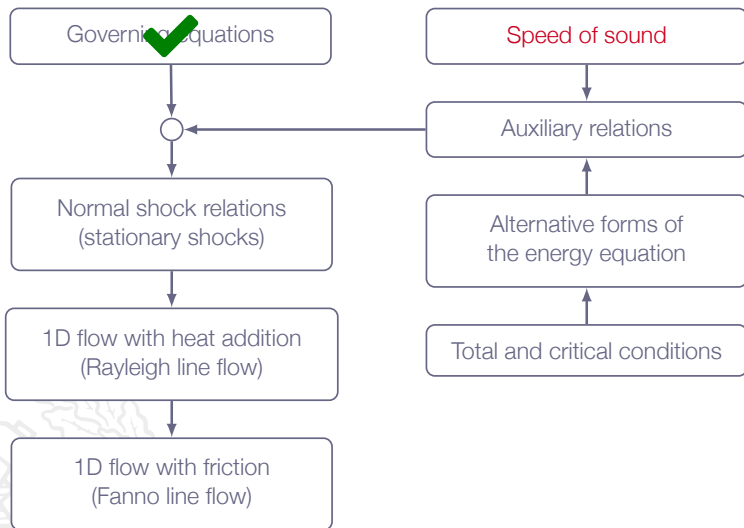
Valid for all gases!

General gas \Rightarrow Numerical solution necessary

Calorically perfect gas \Rightarrow analytical solution exists

Note: These equations are valid regardless of whether there is a shock or not inside the control volume

Roadmap - One-dimensional Flow



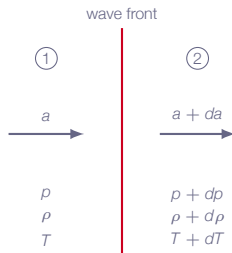
Chapter 3.3

Speed of Sound and Mach Number



Speed of Sound

Sound wave / acoustic perturbation



Speed of Sound

Conservation of mass gives

$$\rho a = (\rho + d\rho)(a + da) = \rho a + \rho da + d\rho a + d\rho da$$

products of infinitesimal quantities are removed \Rightarrow

$$\rho da + d\rho a = 0 \Rightarrow$$

$$a = -\rho \frac{da}{d\rho}$$



Speed of Sound

The momentum equation evaluated over the wave front gives

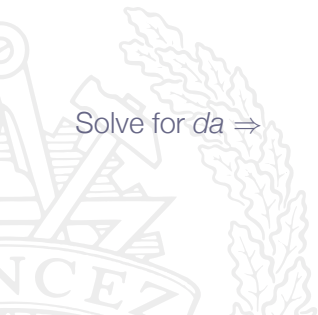
$$\rho + \rho a^2 = (\rho + d\rho) + (\rho + d\rho)(a + da)^2$$

Again, removing products of infinitesimal quantities gives

$$d\rho = -2a\rho da - a^2 d\rho$$

Solve for $da \Rightarrow$

$$da = \frac{d\rho + a^2 d\rho}{-2a\rho}$$



Speed of Sound

Now, inserting the expression for da in the continuity equation gives

$$a = -\rho \left[\frac{dp/d\rho + a^2}{-2a\rho} \right] \Rightarrow$$

$$a^2 = \frac{dp}{d\rho}$$



Speed of Sound

Sound waves are **small perturbations** in ρ , \mathbf{v} , p , T (with constant entropy s) propagating through gas with speed a

$$a^2 = \left(\frac{\partial p}{\partial \rho} \right)_s$$

(valid for all gases)



Speed of Sound

Compressibility and speed of sound:

from before we have

$$\tau_s = \frac{1}{\rho} \left(\frac{\partial \rho}{\partial p} \right)_s$$

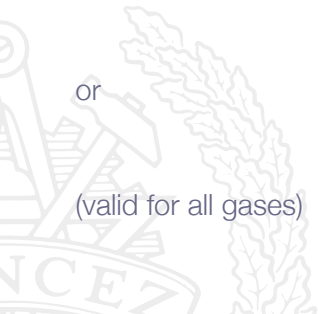
insert in relation for speed of sound

$$a^2 = \left(\frac{\partial p}{\partial \rho} \right)_s = \frac{1}{\rho \tau_s}$$

or

$$a = \sqrt{\frac{1}{\rho \tau_s}}$$

(valid for all gases)



Speed of Sound

Calorically perfect gas:

Isoentropic process $\Rightarrow p = C\rho^\gamma$ (where C is a constant)

$$a^2 = \left(\frac{\partial p}{\partial \rho} \right)_s = \gamma C \rho^{\gamma-1} = \frac{\gamma p}{\rho}$$

which implies

$$a = \sqrt{\frac{\gamma p}{\rho}}$$

or

$$a = \sqrt{\gamma R T}$$



Speed of Sound

Sound wave / acoustic perturbation

- ▶ a **weak wave**
- ▶ propagating through gas at **speed of sound**
- ▶ **small perturbations** in velocity and thermodynamic properties
- ▶ **isentropic** process



Mach Number

The mach number, M , is a local variable

$$M = \frac{v}{a}$$

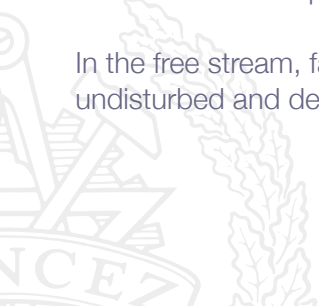
where

$$v = |\mathbf{v}|$$

and a is the local speed of sound

In the free stream, far away from solid objects, the flow is undisturbed and denoted by subscript ∞

$$M_{\infty} = \frac{v_{\infty}}{a_{\infty}}$$



Mach Number

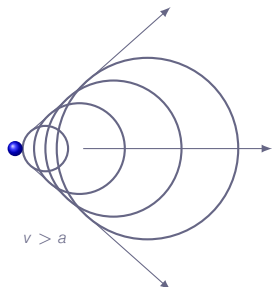
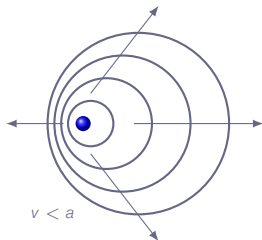
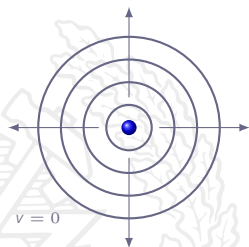
For a fluid element moving along a streamline, the kinetic energy per unit mass and internal energy per unit mass are $V^2/2$ and e , respectively

$$\frac{V^2/2}{e} = \frac{V^2/2}{C_v T} = \frac{V^2/2}{RT/(\gamma - 1)} = \frac{(\gamma/2)V^2}{a^2/(\gamma - 1)} = \frac{\gamma(\gamma - 1)}{2} M^2$$

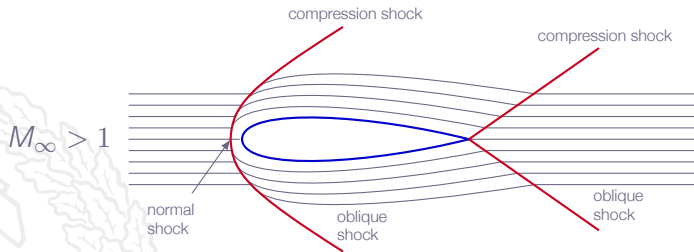
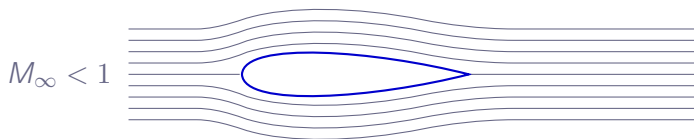
i.e. the Mach number is a measure of the ratio of the **fluid motion** and the **random thermal motion** of the molecules

Physical Consequences of Speed of Sound

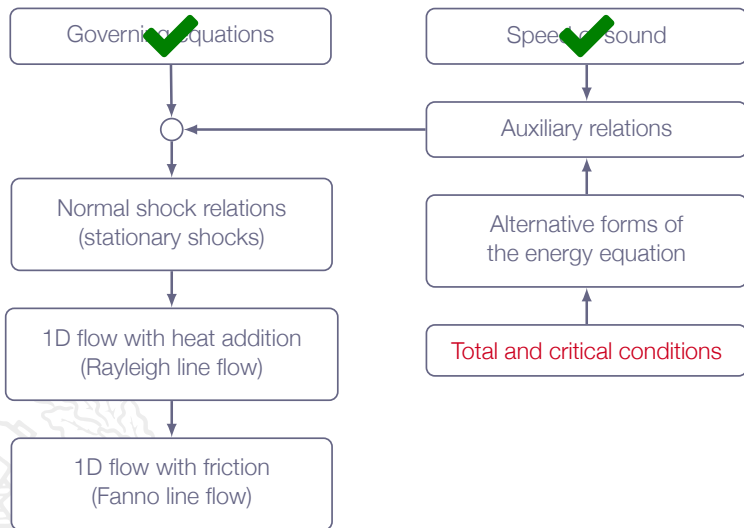
- ▶ Sound waves is the way gas molecules convey information about what is happening in the flow
- ▶ In subsonic flow, sound waves are able to travel upstream, since $v < a$
- ▶ In supersonic flow, sound waves are unable to travel upstream, since $v > a$



Physical Consequences of Speed of Sound



Roadmap - One-dimensional Flow



Chapter 3.4

Some Conveniently Defined Flow Parameters



Stagnation Flow Properties

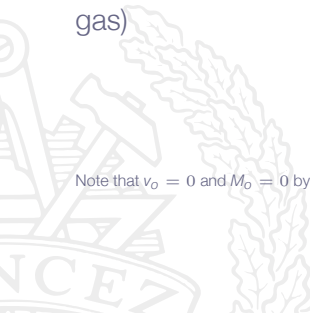
Assumption: Steady inviscid flow

If the flow is **slowed down isentropically** (without flow losses) to **zero velocity** we get the so-called **total conditions** (total pressure p_o , total temperature T_o , total density ρ_o)

Since the process is isentropic, we have (for calorically perfect gas)

$$\frac{p_o}{p} = \left(\frac{\rho_o}{\rho} \right)^\gamma = \left(\frac{T_o}{T} \right)^{\frac{\gamma}{\gamma-1}}$$

Note that $v_o = 0$ and $M_o = 0$ by definition



Critical Conditions

If we **accelerate the flow adiabatically** to the **sonic point**, where $v = a$, we obtain the so-called **critical conditions**, e.g. ρ^* , T^* , ρ^* , a^*

where, by definition, $v^* = a^*$

As for the total conditions, if the process is also reversible (entropy is preserved) we obtain the relations (for calorically perfect gas)

$$\frac{p^*}{p} = \left(\frac{\rho^*}{\rho}\right)^\gamma = \left(\frac{T^*}{T}\right)^{\frac{\gamma}{\gamma-1}}$$



Total and Critical Conditions

For any given steady-state flow and location, we may think of an **imaginary isentropic stagnation process** or an **imaginary adiabatic sonic flow process**

- ▶ We can compute **total** and **critical** conditions at **any point**
- ▶ They represent conditions realizable under an isentropic/adiabatic deceleration or acceleration of the flow
- ▶ Some variables like p_o and T_o will be conserved along streamlines under certain conditions
 - ▶ T_o is conserved along streamlines if the flow is adiabatic
 - ▶ conservation of p_o requires the flow to be isentropic (no viscous losses or shocks)



Total and Critical Conditions

Note: The actual flow does not have to be adiabatic or isentropic from point to point, the total and critical conditions are results of an imaginary isentropic/adiabatic process at one point in the flow.

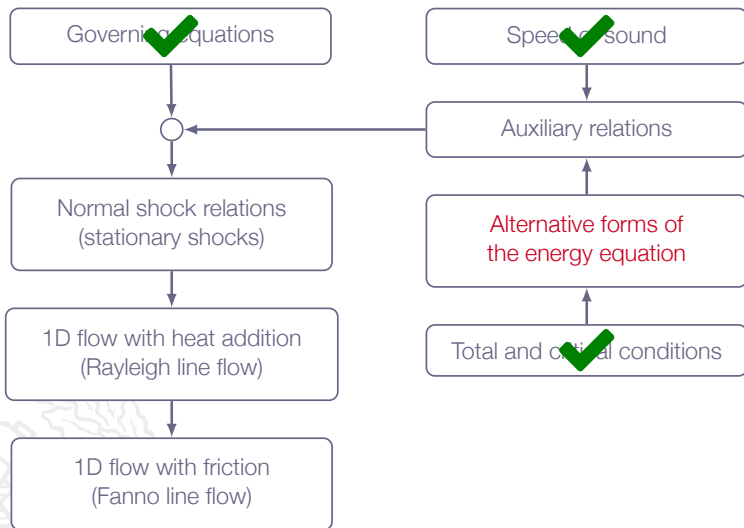
If the flow is not isentropic:

$$T_{0A} \neq T_{0B}, \rho_{0A} \neq \rho_{0B}, \dots$$

However, with isentropic flow T_0, ρ_0, ρ_0 , etc are constants



Roadmap - One-dimensional Flow



Chapter 3.5

Alternative Forms of the Energy Equation



Alternative Forms of the Energy Equation

For steady-state adiabatic flow, we have already shown that conservation of energy gives that total enthalpy, h_o , is constant along streamlines

For a calorically perfect gas we have $h = C_p T$ which implies

$$C_p T + \frac{1}{2} v^2 = C_p T_o$$

$$\frac{T_o}{T} = 1 + \frac{v^2}{2C_p T}$$

Inserting $C_p = \frac{\gamma R}{\gamma - 1}$ and $a^2 = \gamma R T$ we get

$$\frac{T_o}{T} = 1 + \frac{1}{2} (\gamma - 1) M^2$$

Alternative Forms of the Energy Equation

For calorically perfect gas (1D/2D/3D flows):

$$\frac{T_o}{T} = 1 + \frac{1}{2}(\gamma - 1)M^2$$

$$\frac{\rho_o}{\rho} = \left(\frac{T_o}{T}\right)^{\frac{1}{\gamma-1}}$$

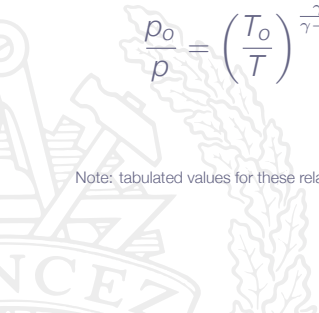
$$\frac{\rho_o}{\rho} = \left(\frac{T_o}{T}\right)^{\frac{\gamma}{\gamma-1}}$$

$$\left(\frac{a^*}{a_o}\right)^2 = \frac{T^*}{T_o} = \frac{2}{\gamma + 1}$$

$$\frac{\rho^*}{\rho_o} = \left(\frac{2}{\gamma + 1}\right)^{\frac{1}{\gamma-1}}$$

$$\frac{p^*}{p_o} = \left(\frac{2}{\gamma + 1}\right)^{\frac{\gamma}{\gamma-1}}$$

Note: tabulated values for these relations can be found in Appendix A.1



Alternative Forms of the Energy Equation

$$M^* \equiv \frac{v}{a^*}$$

For a calorically perfect gas (1D/2D/3D flows)

$$M^2 = \frac{2}{[(\gamma + 1)/M^{*2}] - (\gamma - 1)}$$

This relation between M and M^* gives:

$$M^* = 0 \Leftrightarrow M = 0$$

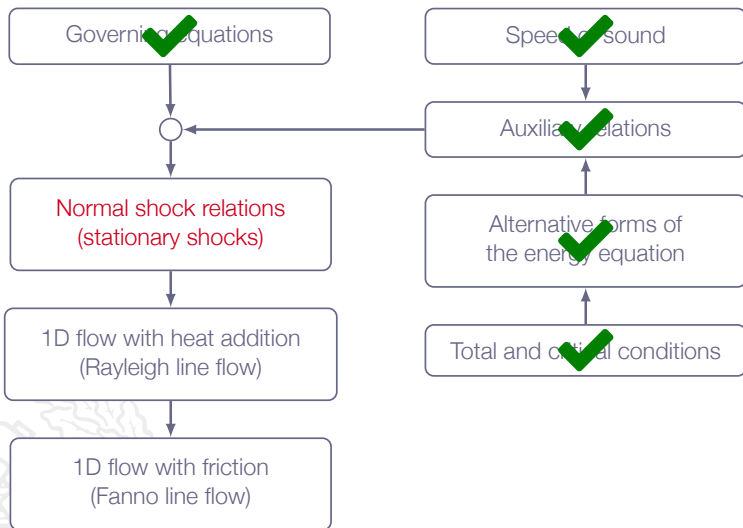
$$M^* = 1 \Leftrightarrow M = 1$$

$$M^* < 1 \Leftrightarrow M < 1$$

$$M^* > 1 \Leftrightarrow M > 1$$

$$M^* \rightarrow \sqrt{\frac{\gamma + 1}{\gamma - 1}} \text{ when } M \rightarrow \infty$$

Roadmap - One-dimensional Flow



Chapter 3.6

Normal Shock Relations



One-Dimensional Flow Equations

$$\rho_1 u_1 = \rho_2 u_2$$

$$\rho_1 u_1^2 + p_1 = \rho_2 u_2^2 + p_2$$

$$h_1 + \frac{1}{2}u_1^2 = h_2 + \frac{1}{2}u_2^2$$



Normal Shock Relations

Calorically perfect gas

$$h = C_p T, \quad p = \rho R T$$

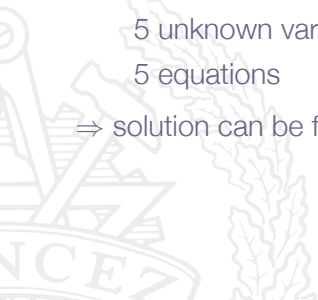
with constant C_p

Assuming that state 1 is known and state 2 is unknown

5 unknown variables: $\rho_2, u_2, p_2, h_2, T_2$

5 equations

⇒ solution can be found



Normal Shock Relations

Divide the momentum equation by $\rho_1 u_1$

$$\frac{1}{\rho_1 u_1} (p_1 + \rho_1 u_1^2) = \frac{1}{\rho_1 u_1} (p_2 + \rho_2 u_2^2)$$

$$\{\rho_1 u_1 = \rho_2 u_2\} \Rightarrow$$

$$\frac{1}{\rho_1 u_1} (p_1 + \rho_1 u_1^2) = \frac{1}{\rho_2 u_2} (p_2 + \rho_2 u_2^2)$$

$$\frac{p_1}{\rho_1 u_1} - \frac{p_2}{\rho_2 u_2} = u_2 - u_1$$



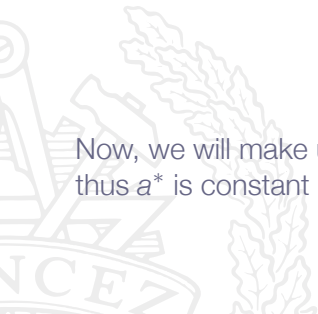
Normal Shock Relations

$$\frac{p_1}{\rho_1 u_1} - \frac{p_2}{\rho_2 u_2} = u_2 - u_1$$

Recall that $a = \sqrt{\frac{\gamma p}{\rho}}$, which gives

$$\frac{a_1^2}{\gamma u_1} - \frac{a_2^2}{\gamma u_2} = u_2 - u_1$$

Now, we will make use of the fact that the flow is adiabatic and thus a^* is constant



Normal Shock Relations

Energy equation:

$$C_p T_1 + \frac{1}{2} u_1^2 = C_p T_2 + \frac{1}{2} u_2^2$$

$$\left\{ C_p = \frac{\gamma R}{\gamma - 1} \right\} \Rightarrow$$

$$\frac{\gamma R T_1}{(\gamma - 1)} + \frac{1}{2} u_1^2 = \frac{\gamma R T_2}{(\gamma - 1)} + \frac{1}{2} u_2^2$$

$$\left\{ a = \sqrt{\gamma R T} \right\} \Rightarrow$$

$$\frac{a_1^2}{(\gamma - 1)} + \frac{1}{2} u_1^2 = \frac{a_2^2}{(\gamma - 1)} + \frac{1}{2} u_2^2$$



Normal Shock Relations

In any position in the flow we can get a relation between the local speed of sound a , the local velocity u , and the speed of sound at sonic conditions a^* by inserting in the equation on the previous slide. $u_1 = u, a_1 = a, u_2 = a_2 = a^* \Rightarrow$

$$\frac{a^2}{(\gamma - 1)} + \frac{1}{2}u^2 = \frac{a^{*2}}{(\gamma - 1)} + \frac{1}{2}a^{*2}$$

$$a^2 = \frac{\gamma + 1}{2}a^{*2} - \frac{\gamma - 1}{2}u^2$$

Evaluated in station 1 and 2, this gives

$$a_1^2 = \frac{\gamma + 1}{2}a^{*2} - \frac{\gamma - 1}{2}u_1^2$$

$$a_2^2 = \frac{\gamma + 1}{2}a^{*2} - \frac{\gamma - 1}{2}u_2^2$$

Normal Shock Relations

Now, inserting

$$a_1^2 = \frac{\gamma + 1}{2} a^{*2} - \frac{\gamma - 1}{2} u_1^2$$

and

$$a_2^2 = \frac{\gamma + 1}{2} a^{*2} - \frac{\gamma - 1}{2} u_2^2$$

in

$$\frac{a_1^2}{(\gamma - 1)} + \frac{1}{2} u_1^2 = \frac{a_2^2}{(\gamma - 1)} + \frac{1}{2} u_2^2$$

and solve for a^* gives

$$a^{*2} = u_1 u_2$$

Normal Shock Relations

$$a^{*2} = u_1 u_2$$

A.K.A. the Prandtl relation. Divide by a^{*2} on both sides \Rightarrow

$$1 = \frac{u_1}{a^*} \frac{u_2}{a^*} = M_1^* M_2^*$$

Together with the relation between M and M^* , this gives

$$M_2^2 = \frac{1 + \frac{1}{2}(\gamma - 1)M_1^2}{\gamma M_1^2 - \frac{1}{2}(\gamma - 1)}$$

Normal Shock Relations

Continuity equation and $a^{*2} = u_1 u_2$

$$\frac{\rho_2}{\rho_1} = \frac{u_1}{u_2} = \frac{u_1^2}{u_1 u_2} = \frac{u_1^2}{a^{*2}} = M_1^*$$

which gives

$$\frac{\rho_2}{\rho_1} = \frac{u_1}{u_2} = \frac{(\gamma + 1)M_1}{2 + (\gamma - 1)M_1}$$



Normal Shock Relations

Now, once again back to the momentum equation

$$p_2 - p_1 = \rho_1 u_1^2 - \rho_2 u_2^2 = \{\rho_1 u_1 = \rho_2 u_2\} = \rho_1 u_1 (u_1 - u_2)$$

$$\frac{p_2}{p_1} - 1 = \frac{\rho_1 u_1^2}{\rho_1} \left(1 - \frac{u_2}{u_1}\right) = \left\{a = \sqrt{\frac{\gamma p}{\rho}}\right\} = \gamma M_1^2 \left(1 - \frac{u_2}{u_1}\right)$$

with the expression for u_2/u_1 derived previously, this gives

$$\frac{p_2}{p_1} = 1 + \frac{2\gamma}{\gamma + 1} (M_1^2 - 1)$$

Normal Shock Relations

Normal shock $\Rightarrow M_1 > 1$

$$M_1 > 1 \Rightarrow M_1^* > 1$$

$$M_2^* = \frac{1}{M_1^*} \Rightarrow M_2^* < 1$$

$$M_2^* < 1 \Rightarrow M_2 < 1$$

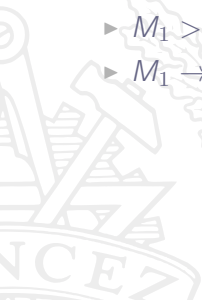
After a normal shock the Mach number must be lower than 1.0



Normal Shock Relations

$$M_2^2 = \frac{1 + \frac{1}{2}(\gamma - 1)M_1^2}{\gamma M_1^2 - \frac{1}{2}(\gamma - 1)}$$

- ▶ $M_1 = 1.0 \Rightarrow M_2 = 1.0$
- ▶ $M_1 > 1.0 \Rightarrow M_2 < 1.0$
- ▶ $M_1 \rightarrow \infty \Rightarrow M_2 \rightarrow \sqrt{(\gamma - 1)/(2\gamma)} = \{\gamma = 1.4\} = 0.378$



Normal Shock Relations

Are the normal shock relations valid for $M_1 < 1.0$?

Mathematically - yes!

Physically - ?



Normal Shock Relations

Let's have a look at the 2nd law of thermodynamics

$$s_2 - s_1 = C_p \ln \frac{T_2}{T_1} - R \ln \frac{p_2}{p_1}$$

We get the ratios (T_2/T_1) and (p_2/p_1) from the normal shock relations

$$s_2 - s_1 = C_p \ln \left[\left(1 + \frac{2\gamma}{\gamma+1} (M_1^2 - 1) \right) \left(\frac{2 + (\gamma-1)M_1^2}{(\gamma+1)M_1^2} \right) \right] + \\ - R \ln \left(1 + \frac{2\gamma}{\gamma+1} (M_1^2 - 1) \right)$$

$$M_1 = 1 \Rightarrow \Delta s = 0 \text{ (Mach wave)}$$

$$M_1 < 1 \Rightarrow \Delta s < 0 \text{ (not physical)}$$

$$M_1 > 1 \Rightarrow \Delta s > 0$$

Normal Shock Relations

$M_1 > 1.0$ gives $M_2 < 1.0$, $\rho_2 > \rho_1$, $p_2 > p_1$, and $T_2 > T_1$

What about T_o and p_o ?

Energy equation:

$$C_p T_1 + \frac{u_1^2}{2} = C_p T_2 + \frac{u_2^2}{2}$$

$$C_p T_{o1} = C_p T_{o2}$$

calorically perfect gas \Rightarrow

$$T_{o1} = T_{o2}$$

or more general (as long as the shock is stationary):

$$h_{o1} = h_{o2}$$

Normal Shock Relations

2nd law of thermodynamics and isentropic deceleration to zero velocity (Δs unchanged since isentropic) gives

$$s_2 - s_1 = C_p \ln \frac{T_{o2}}{T_{o1}} - R \ln \frac{\rho_{o2}}{\rho_{o1}} = \{T_{o1} = T_{o2}\} = -R \ln \frac{\rho_{o2}}{\rho_{o1}}$$

$$\frac{\rho_{o2}}{\rho_{o1}} = e^{-(s_2 - s_1)/R}$$

i.e. the total pressure decreases over a normal shock



Normal Shock Relations

Normal shock relations for calorically perfect gas (summary):

$$T_{o1} = T_{o2}$$

$$a_{o1} = a_{o2}$$

$$a_1^* = a_2^* = a^*$$

$$u_1 u_2 = a^{*2}$$

(the Prandtl relation)

$$M_2^* = \frac{1}{M_1^*}$$

$$M_2^2 = \frac{1 + \frac{1}{2}(\gamma - 1)M_1^2}{\gamma M_1^2 - \frac{1}{2}(\gamma - 1)}$$

$$\frac{\rho_2}{\rho_1} = 1 + \frac{2\gamma}{\gamma + 1}(M_1^2 - 1)$$

$$\frac{\rho_2}{\rho_1} = \frac{u_1}{u_2} = \frac{(\gamma + 1)M_1^2}{2 + (\gamma - 1)M_1^2}$$

$$\frac{T_2}{T_1} = \frac{\rho_2}{\rho_1} \frac{\rho_1}{\rho_2}$$

see table A.2 and figure 3.10 on p. 94

Normal Shock Relations

As the flow passes a stationary normal shock, the following changes will take place discontinuously across the shock:

ρ increases

ρ increases

u decreases

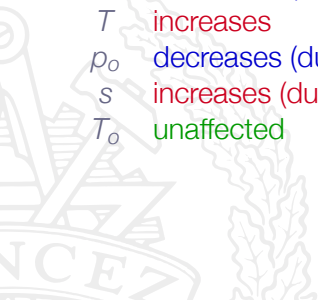
M decreases (from $M > 1$ to $M < 1$)

T increases

ρ_o decreases (due to shock loss)

s increases (due to shock loss)

T_o unaffected



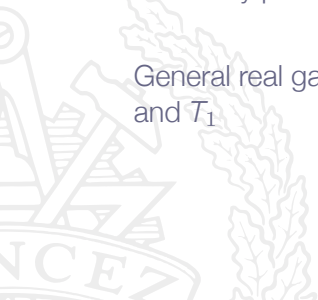
Normal Shock Relations

The normal shock relations for calorically perfect gases are valid for $M_1 \leq 5$ (approximately) for air at standard conditions

Calorically perfect gas \Rightarrow Shock depends on M_1 only

Thermally perfect gas \Rightarrow Shock depends on M_1 and T_1

General real gas (non-perfect) \Rightarrow Shock depends on $M_1, p_1,$
and T_1



Normal Shock Relations

And now to the question that probably bothers most of you
but that no one dares to ask ...



Normal Shock Relations

And now to the question that probably bothers most of you but that no one dares to ask ...

When or where did we say that there was going to be a shock between 1 and 2?



Normal Shock Relations

And now to the question that probably bothers most of you but that no one dares to ask ...

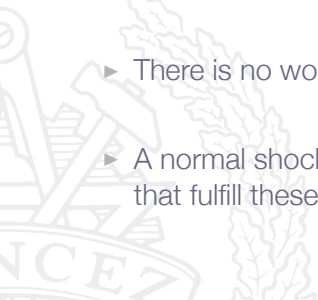
When or where did we say that there was going to be a shock between 1 and 2?

Answer: We did not (explicitly)



Normal Shock Relations

- ▶ The derivation is based on the fact that there should be a change in flow properties between 1 and 2
- ▶ We are assuming steady state conditions
- ▶ We have said that the flow is adiabatic (no added or removed heat)
- ▶ There is no work done and no friction added
- ▶ A normal shock is the solution provided by nature (and math) that fulfill these requirements!



Chapter 3.7

Hugoniot Equation



Hugoniot Equation

Starting point: governing equations for normal shocks

$$\rho_1 u_1 = \rho_2 u_2$$

$$\rho_1 u_1^2 + p_1 = \rho_2 u_2^2 + p_2$$

$$h_1 + \frac{1}{2}u_1^2 = h_2 + \frac{1}{2}u_2^2$$

Eliminate u_1 and u_2 gives:

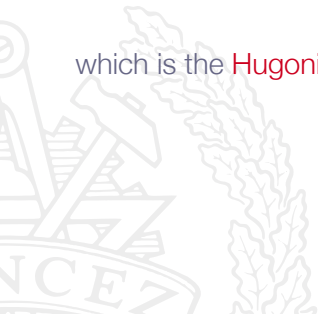
$$h_2 - h_1 = \frac{\rho_2 - \rho_1}{2} \left(\frac{1}{\rho_1} + \frac{1}{\rho_2} \right)$$

Hugoniot Equation

Now, insert $h = e + p/\rho$ gives

$$e_2 - e_1 = \frac{\rho_2 + \rho_1}{2} \left(\frac{1}{\rho_1} - \frac{1}{\rho_2} \right) = \frac{\rho_2 + \rho_1}{2} (\nu_1 - \nu_2)$$

which is the **Hugoniot relation**



Stationary Normal Shock in One-Dimensional Flow

Normal shock:

$$e_2 - e_1 = -\frac{\rho_2 + \rho_1}{2} (\nu_2 - \nu_1)$$

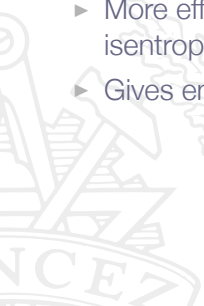
- ▶ More effective than isentropic process
- ▶ Gives entropy increase

Isentropic process:

$$de = -pd\nu$$

- ▶ More efficient than normal shock process

see figure 3.11 p. 100



Roadmap - One-dimensional Flow

