Cite as: Barba Piña, J.L.A.: Implementation of a mass flux term with thermodiffusion mass transport into the species transport equation in a compressible solver. In Proceedings of CFD with OpenSource Software, 2019, Edited by Nilsson. H., http://dx.doi.org/10.17196/OS_CFD#YEAR_2019

CFD WITH OPENSOURCE SOFTWARE

A COURSE AT CHALMERS UNIVERSITY OF TECHNOLOGY TAUGHT BY HÅKAN NILSSON

Implementation of a mass flux term with thermodiffusion mass transport into the species transport equation in a compressible solver

Developed for OpenFOAM-v1906

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January 16, 2020

Learning outcomes

The reader will learn: How to use it:

- How thermophysical properties are structured through the thermophysicalModels library and how the thermophysicalProperties dictionary works for compressible solvers.
- How to implement a species transport equation for binary gas mixtures into a compressible solver.

The theory of it:

- A description of the thermophysical properties in OpenFOAM context and how thermophysical models are structured under the *thermophysicalProperties* dictionary entries.
- The theory of thermo-diffusion separation phenomena in fluid mixtures (also known as Ludwig-Soret effect).
- The definition of constant mass transport coefficients and how they are defined and accessed through the *thermophysicalProperties* dictionary.

How it is implemented:

- The implementation of new transport coefficients by modifying the *consTransport* class which is part of the *specie thermophysicalModels* sub-library.
- The addition of the new transport model and the new added transport coefficients into a main thermophysical model, based on *rhoThermo* model contained in the *basic thermophysicalModels* sub-library.
- The addition of the species transport equation for binary gas mixtures for a compressible solver (*rhoSimpleFoam*) which will work with the newly added transport coefficients.

How to modify it:

- Through the modification of the *thermophysicalModels* library so it can include the mass diffusion and thermodiffusion coefficients.
- The new transport coefficients should be accessed trhough the thermophysicalProperties library so they can be added into the species transport equation for binary gas mixtures.
- A new compressible solver capable of solving the species transport equation should be tested by a simple tutorial case, so mass transfer by thermodiffusion can be visualized.

Prerequisites

The reader is expected to know the following in order to get maximum benefit out of this report:

- Basic knowledge on heat and mass transfer processes (check reference [2]).
- A general knowledge of OpenFOAM's code organization, as well as some knowledge of C++ object-oriented programming.
- It is recommended that the reader gets a general overview on thermodiffusion phenomena in gas mixtures. References [1, 4, 5, 8, 9] can be consulted for this purpose.
- As a suggestion, if there is an extra interest in how to implement or change OpenFOAM's *thermophysicalModels* library for different applications, the reader can consult the works from Hummel, D. [6] and Choquet, I. [3].

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Chapter 1

Theoretical background

1.1 Introduction

In separation processes, a general approach for getting purified products most efficiently is to take advantage of some transport properties of the mixture system. In the specific case of homogeneous fluid mixtures, there is an interesting phenomenon that can be used for the purification of mixture components, that is, by inducing temperature gradients in the flow field which combined with other molecular transport and diffusive processes can produce a total mass flux of each component. This process is known as thermophoresis, thermodiffusion, or the Ludwig-Soret effect for liquid mixtures [9]. There are several natural and industrial processes where thermophoresis takes part as the main driving force for mass diffusivity, such phenomena includes the thermohaline circulation which is of highly importance in large scale ocean circulation [4], and thermogravitational columns used for separation of gaseous mixtures, crude oil and mixtures of liquid polymers [4, 8, 9].

It would be interesting to analize this kind of processes in the context of CFD since the involvement of energy and momentum transport affects the behavior of mixture separation, which makes the problem complex enough to be tackled by numerical analysis. Luckily, OpenFOAM already counts with a structured library that allows the simulation of cases where energy transfer can be modeled through the coupling of thermodynamic, transport, and specie interaction models.

The purpose of this tutorial is to contribute with a basic numerical model that can be used for the study of the thermodiffusion separation in binary gas mixtures, showing a method for code modification and addition in OpenFOAM's libraries. The writer hopes that this method can help the interested reader to be used for similar implementations.

1.2 Mass diffusion caused by temperature gradients

Temperature gradients in fluid homogeneous mixtures results in a relative concentration difference depending on the temperature field values, at the same time, regular mass diffusion is caused by these concentration gradients. Terefore, it can be said that in a temperature steady-state system, the total mass flux equals to a balance between the ordinary mass diffusion and thermodiffusion transport. For gas mixtures, a theoretical description of this mechanism was made by Chapman and Enskog based on the kinetic theory of gases [1]. The mathematical expression for the net mass flux for a reference component a of a binary gas mixture, with constant pressure and without the effect of external forces, can be expressed as in Rahman et al. [9] by the following equation,

$$J_a(x, y, z, C_a, T, \rho) = \rho \left[D_{ab} \nabla C_a + D_T C_{a0} (1 - C_{a0}) \nabla T \right]$$
(1.1)

Where $J_a(x, y, z, C_a, T, \rho)$ is the net mass flux of component *a* which is a function of temperature, concentration, density, and the direction of the flow (in this case, represented by the cartesian space

coordinates of the system: x, y, z). Here, $\rho = f(p, T)$ is the density of the mixture, D_{ab} is the binary mass diffusion coefficient $[\frac{m}{s^2}]$, C_a is the mass fraction of the reference component a, C_{a0} is the initial mass fraction of component a, and D_T is the thermodiffusion coefficient $[\frac{m}{K \cdot s^2}]$. In a steady state system, the mass flux is at equilibrium, so that $J_a = 0$ and the concentration gradient can be set on terms of the temperature gradient and the ratio of the transport coefficients

$$\nabla C_a = -\frac{D_T}{D_{ab}} C_{a0} (1 - C_{a0}) \nabla T \tag{1.2}$$

Dimensionless quantities can help to obtain the values of the transport coefficients needed in the calculation of transport equations such as in equation (1.1). This feature can be used for getting unknown physical properties from other known ones, which is actually done by OpenFOAM's constTransport class for calculating the thermal conductivity for simulation cases that use the thermophysicalProperties library (Something that will be used for this tutorial).

The thermodiffusion coefficient D_T can be obtained with a dimensionless number known as thermodiffusion ratio K_T [5]. This dimensionless quantity represents the relative effects between thermodiffusion and ordinary mass diffusion at average temperature

$$K_T = T \frac{D_T}{D_{ab}} \tag{1.3}$$

Like the thermodiffusion and mass diffusivity transport coefficients, the value of K_T depends on several molecular parameters, such as molecular masses, the molecular size, the mixture composition, temperature and intermolecular interactions [1]. Regardless of the complexity of all of these variable interactions, a simple interpretation of the value of this coefficient can be given, when K_T is positive, the heaviest specie molecules tend to move toward a colder region, and when it is negative, the heaviest molecules move toward a warmer region [1]. Equation (1.1) can be rearranged in terms of K_T

$$J_a = \rho D_{ab} \left(\nabla C_a + K_T \frac{C_{a0}(1 - C_{a0})}{T} \nabla T \right)$$
(1.4)

If the temperature and concentration gradients are assumed to depend on one dimension only (let's say x), for a steady state process, the equation (1.2) can be expressed as follows

$$\frac{\partial C_a}{\partial x} = -K_T \frac{C_{a0}(1 - C_{a0})}{T} \frac{\partial T}{\partial x}$$
(1.5)

If thermodiffusion ratio K_T is taken as a constant value independent of the composition of the mixture, an integration can be carried out between temperatures T_c and T_h (lower to higher temperature), giving the following expression

$$\Delta C_a = -K_T C_{a0} (1 - C_{a0}) \ln \left(\frac{T_h}{T_c}\right) \tag{1.6}$$

This equation, can be used for validating the thermodiffusion simulation case that will be used to test the code implementations shown in the following chapters.

1.3 Thermophysical models in OpenFOAM

When simulating flow problems where energy and mass transfer occur in compressible systems it is necessary to implement a model that couples the requirement of a state equation, thermodynamic fluid properties, transport properties, and energy calculation models. For this kind of problems, OpenFOAM already counts with a broad library that allows retrieving thermophysical properties as constant values or functions of temperature, pressure, and composition, with an additional set of equations of state that all together can be used to obtain thermal energy calculations in terms of enthalpy or internal energy ([3]).

The *thermophysicalModels* library is constituted as a set of models that allows calculating several fluid properties that depend on temperature, pressure, and composition of a fluid (or solid) mixture. The structure of this library is organized in such a way that the main thermophysical models depend on other submodels for mixture properties, which in turn depend on transport, thermodynamic properties, and state equations submodels. Table 1.1 shows which thermophysical variables correspond to each submodel, this dependence forms a complex built library that allows the simulation of different types of problems such as compressible flows, heat transfer, multiphase flows, combustion, etc.

Thermophysical Model Structure						
Submodel	Variables and fluid properties					
Mixture models	$T, p, X_a, C_a, $ etc.					
Transport models	μ , κ , $lpha$, etc.					
Thermodynamic properties	$C_p, C_v, H_f, e, $ etc.					
Equations of state	ρ calculations					

Table 1.1: Basic structure of the thermophysical modelling in OpenFOAM.

OpenFOAM counts with submodels sets that are combined to form a main thermophysical model forming the first *layer* of the model to be used, then, a second layer is necessary to define the mixture type, which is formed with the *specie*, *thermodynamics* and *transport* numerical value entries. As a reference, the table 1.2 contains a list of each set of the thermophysical submodels ([7]).

1-Equation of State - equation Of State	Description	
icoPolvnomial	Incompressible polynomial equation of state, e.g. for	
porfectC as	liquids Perfect gas equation of state	
2-Basic thermophysical properties — thermo	Description	
eConstThermo	Constant specific heat cp model with evaluation of	
hConstThermo	internal energy e and entropy s Constant specific heat cp model with evaluation of enthalpy h and entropy s cp evaluated by a function with coefficients from polynomi- als, from which h , s are evaluated C_{-} evaluated by a function with coefficients from	
hPolynomialThermo		
janafThermo	JANAF thermodynamic tables, from which h , s are evaluated	
3-Derived thermophysical properties — specieThermo	Description	
specieThermo	Thermophysical properties of species, derived from C_p , h and/or s	
4-Transport properties — transport	Description	
$\operatorname{const}\operatorname{Transport}$	Constant transport properties	
polynomialTransport	Polynomial based temperature-dependent transport properties	
${ m sutherland}{ m Transport}$	Sutherland's formula for temperature-dependent transport properties	
5-Mixture properties — mixture	Description	
pureMixture	General thermophysical model calculation for passive	
${ m homogeneous}{ m Mixture}$	gas mixtures Combustion mixture based on normalised fuel mass fraction b	
inhomogeneousMixture	Combustion mixture based on b and total fuel mass fraction f_t	
very Inhomogeneous Mixture	Combustion mixture based on b, ft and unburnt fuel mass fraction f_{a}	
${ m diesel}{ m Mixt}{ m ure}$	Combustion mixture based on f_t and f_u	
${\it basicMultiComponentMixture}$	Basic mixture based on multiple components	
${ m multiComponent}{ m Mixture}$	Derived mixture based on multiple components	
reactingMixture	Combustion mixture using thermodynamics and re- action schemes	
egrMixture	Exhaust gas recirc ulation mixture	
6-Thermophysical model — thermoModel	Description	
hePsiThermo	General thermophysical model calculation based on enthalpy h or internal energy e , and compressibility Φ	
heRhoThermo	General thermophysical model calculation based on enthalpy h or internal energy e , and density ρ	
${ m hePsiMixtureThermo}$	Calculates enthalpy for combustion mixture based on enthalpy h or internal energy e and Φ	
${ m heRhoMixtureThermo}$	Calculates enthalpy for combustion mixture based on enthalpy h or internal energy e , and ρ	
${\rm heheuMixtureThermo}$	Calculates enthalpy h or internal energy e for un- burnt u gas and combustion mixture	

untion fC tion Of St F.

Table 1.2: Submodel sets of the thermophysical Models library, (retrieved from OpenFOAM User Guide [7])

1.3.1 thermophysicalProperties dictionary

The *thermophysicalProperties* dictionary is the file where the user can specify the entry values for any solver that uses the *thermophysicalModels* library. In a simulation case, this file can be found in the *constant* folder. The structure of this file begins with the chosen thermophysical model, formed by a combination of each of the thermophysical properties submodels that are specified in the two entry *layers* mentioned in the previous section. The following example of a *thermophysicalProperties* dictionary gives a brief explanation of how this dictionary is constituted.

```
FoamFile
{
version
           2.0;
format
           ascii;
class
           dictionary;
location
           "constant";
           thermophysicalProperties;
object
}
//- First layer of the thermophysical model, submodels from table 1.2
// are specified here
thermoType
{
                   hePsiThermo;
    type
                   pureMixture;
    mixture
    transport
                   const;
    thermo
                   eConst;
    equationOfState perfectGas;
    specie
                    specie;
    energy
                   sensibleInternalEnergy;
}
//- Second layer of the thermophysical model, the model is constructed with the entry
// values specified here, starting with the "mixture" keyword that involves
// the "specie" , "thermodynamics", and "transport" entries, corresponding with the
// submodel settings in the first layer.
mixture
{
specie
{
              28.9;
  molWeight
}
thermodynamics
{
  Cv
              712;
  Hf
              0;
}
transport
{
              1.8e-05;
  mu
  Pr
              0.7;
}
}
// *
                                                         * * * * * * * * //
                                  * * * * *
                                *
```

It is important to mention that not all the submodel combinations are allowed. Checking the possible combinations can be done by copying a tutorial case that uses thermophysical modelling, (e.g. $$WM_PROJECT_DIR/tutorials/compressible/rhoPimpleFoam/laminar/sineWaveDamping$) and then try to change one of the entries in the thermoPhysicalProperties dictionary (e.g. change const for banana). Now, when running the case, the output will show an error message complaning that the combination is not appropriate and will show a list of the possible thermophysicalProperties combinations. An example of this error message can be seen as follows,

```
Create time
Create mesh for time = 68
SIMPLE: convergence criteria
             tolerance 0.001
field p
field U
             tolerance 0.0001
field e
             tolerance 0.001
Reading thermophysical properties
Selecting thermodynamics package
                heRhoThermo:
type
                pureMixture;
mixture
transport
                banana:
thermo
                hConst;
equationOfState perfectGas;
specie
                specie;
energy
                sensibleEnthalpy;
3
--> FOAM FATAL ERROR:
Unknown fluidThermo type
thermoType
ł
                heRhoThermo;
type
                pureMixture;
mixture
transport
                banana;
                hConst;
thermo
equationOfState perfectGas;
                specie;
specie
                sensibleEnthalpy;
energy
Valid fluidThermo types are:
                                                 hConst
                                                          perfectGas
                                                                        specie sensibleEnthalpy
hePsiThermo
             homogeneousMixture
                                     const
hePsiThermo
              homogeneousMixture
                                                 hConst
                                                          perfectGas
                                                                                sensibleEnthalpy
                                     sutherland
                                                                        specie
hePsiThermo
              homogeneousMixture
                                     sutherland
                                                 janaf
                                                          perfectGas
                                                                        specie
                                                                                 sensibleEnthalpy
              inhomogeneousMixture
                                                 hConst
                                                                        specie
                                                                                sensibleEnthalpy
hePsiThermo
                                                          perfectGas
                                     const
hePsiThermo
              inhomogeneousMixture
                                     sutherland
                                                 hConst
                                                          perfectGas
                                                                                 sensibleEnthalpy
                                                                        specie
              inhomogeneousMixture
                                                          perfectGas
hePsiThermo
                                     sutherland
                                                                                sensibleEnthalpy
                                                ianaf
                                                                        specie
hePsiThermo
              multiComponentMixture
                                     const
                                                 eConst
                                                          perfectGas
                                                                        specie
                                                                                sensibleInternalEnergy
. . . . . . .
heRhoThermo homogeneousMixture
                                   const
                                             hConst incompressiblePerfectGas specie sensibleEnthalpy
                                                                        specie sensibleEnthalpy
             homogeneousMixture
                                             hConst perfectGas
heRhoThermo
                                   const
heRhoThermo
             homogeneousMixture
                                   sutherland
                                                janaf incompressiblePerfectGas specie sensibleEnthalpy
heRhoThermo
             homogeneousMixture
                                   sutherland
                                                janaf perfectGas
                                                                        specie sensibleEnthalpy
                                             hConst incompressiblePerfectGas
heRhoThermo
             inhomogeneousMixture
                                   const
                                                                                 specie sensibleEnthalpv
                                                                        specie sensibleEnthalpy
heRhoThermo
             inhomogeneousMixture
                                   const
                                             hConst perfectGas
heRhoThermo
             inhomogeneousMixture
                                   sutherland
                                                janaf incompressiblePerfectGas specie sensibleEnthalpy
             inhomogeneousMixture
heRhoThermo
                                   sutherland
                                                janaf perfectGas
                                                                        specie sensibleEnthalpy
heRhoThermo
             multiComponentMixture const
                                             eConst adiabaticPerfectFluid
                                                                               specie sensibleInternalEnergy
heRhoThermo
             multiComponentMixture const
                                             eConst
                                                     incompressiblePerfectGas specie sensibleInternalEnergy
heRhoThermo
             multiComponentMixture const
                                             eConst perfectFluid
                                                                      specie sensibleInternalEnergy
```

```
... ...
```

Chapter 2

Implementing constant transport coefficients through thermophysicalProperties library

2.1 Creating the *thermophysicalModels* user library

Before doing any new implementation, the first thing to do is to make sure that the original code will not be affected, this is done by copying the required library folders and substituting the original compilation path with the users one. The following instructions are meant to show how to do this for the *specie* and *basic* sub-libraries which are part of the thermophysicalModels library.

2.1.1 specie and basic libraries

The *specie* library contains models for transport properties, state equations, thermodynamics and chemical reactions. In a terminal window with the *OpenFOAM-v1906* enviriment active, copy the following lines to get your own version of the *specie* library.

mkdir \$WM_PROJECT_USER_DIR/src/thermophysicalModels
cd \$WM_PROJECT_USER_DIR/src/thermophysicalModels
cp -r \$FOAM_SRC/thermophysicalModels/specie .
sed -i s/"FOAM_LIBBIN"/"FOAM_USER_LIBBIN"/g specie/Make/files

The thermophysical base models are defined in the *basic* library, the type of calculations depends on the base classes that are contained here (e.g. rhoThermo or psiThermo). Copy the following lines to get your own version of the *basic* library. Notice that the name of the *specie* and *basic* libraries are not changed, it helps to avoid renaming all the file dependencies besides the new *thermophysicalModel* library coexist in parallel with the original one. Additionally, the line -L\$(FOAM_USER_LIBBIN) gives priority to the user's library when the compiler looks for the linked files [3].

```
cp -r $FOAM_SRC/thermophysicalModels/basic $WM_PROJECT_USER_DIR/src/thermophysicalModels
cd $WM_PROJECT_USER_DIR/src/thermophysicalModels/basic
sed -i s/"FOAM_LIBBIN"/"FOAM_USER_LIBBIN"/g ./Make/files
sed -i '/EXE_INC = \\/a -I$(WM_PROJECT_USER_DIR)/src/thermophysicalModels/specie/lnInclude \\' ./Make/options
sed -i '/LIB_LIBS = \\/a -L$(FOAM_USER_LIBBIN) \\' ./Make/options
```

Now, let's check that the *sed* commands have worked properly for the *files* and *options* files in both libraries. We open the *specie* library *files* file just to check that the library code will be compiled in the user's library directory:

• • •

```
atomicWeights/atomicWeights.C
specie/specie.C
reaction/reactions/makeReactions.C
reaction/reactions/makeLangmuirHinshelwoodReactions.C
```

LIB = \$(FOAM_USER_LIBBIN)/libspecie

• • •

Now, we continue with the *specie* library *options* file and we verify that no executable or library paths are written in this file:

```
EXE_INC =
```

Then, we check that the *files* file of the *basic* library looks as follows:

```
basicThermo/basicThermo.C
fluidThermo/fluidThermo.C
psiThermo/psiThermo.C
psiThermo/psiThermo.C
rhoThermo/rhoThermo.C
rhoThermo/liquidThermo.C
derivedFvPatchFields/fixedEnergy/fixedEnergyFvPatchScalarField.C
derivedFvPatchFields/gradientEnergy/gradientEnergyFvPatchScalarField.C
derivedFvPatchFields/mixedEnergy/mixedEnergyFvPatchScalarField.C
derivedFvPatchFields/mixedEnergy/mixedEnergyFvPatchScalarField.C
derivedFvPatchFields/energyJump/energyJumpFvPatchScalarField.C
derivedFvPatchFields/energyJump/energyJumpFvPatchScalarField.C
derivedFvPatchFields/energyJump/energyJumpAMI/energyJumpAMIFvPatchScalarField.C
LIB = $(FOAM_USER_LIBBIN)/libfluidThermophysicalModels
```

. . .

And finally, we check on the *options* file of the *basic* library:

```
EXE_INC = \
    -I$(WM_PROJECT_USER_DIR)/src/thermophysicalModels/specie/lnInclude \
    -I$(LIB_SRC)/finiteVolume/lnInclude \
    -I$(LIB_SRC)/meshTools/lnInclude \
    -I$(LIB_SRC)/transportModels/compressible/lnInclude \
    -I$(LIB_SRC)/thermophysicalModels/specie/lnInclude \
    -I$(LIB_SRC)/thermophysicalModels/specie/lnInclude \
    -I$(LIB_SRC)/thermophysicalModels/thermophysicalProperties/lnInclude \
    LIB_LIBS = \
    -L$(FOAM_USER_LIBBIN) \
    -lfiniteVolume \
    -ImeshTools \
```

```
-lcompressibleTransportModels \
-lspecie \
-lthermophysicalProperties
```

. . .

Now, we can proceed with the compilation of the new libraries, beginning with the *specie* library, and then continuing with the *basic* library. We start with the *specie* library compilation by writing the next command lines in the terminal window.

```
cd $WM_PROJECT_USER_DIR/src/thermophysicalModels
wclean lib specie
wmake libso specie
```

If the compilation process for the *specie* library finishes without errors, we can continue with the compilation of the *basic* library.

```
cd $WM_PROJECT_USER_DIR/src/thermophysicalModels
wclean lib basic
wmake libso basic
```

2.2 Creating the new transport model: thermoDiffconstTransport

At the user's ./thermophysicalModels/specie directory, one can find the transport models folder where the const transport code model for constant transport properties is located. In this model, thermal conductivity and thermal diffusivity are calculated from thermophysicalProperties dictionary entries mu and Pr (dynamic viscosity and Prandtl number respectively).

The idea is to use the *const* transport model as a base for our new *thermoDiffusionTransport* model, where we will define two entries Dab and KT, which are the mass diffusivity and the thermodiffusion coefficient that are going to be used for calculating the thermodiffusion coefficient DT.

The first step is to copy the *constTransport* and rename the copied folder and the files inside it. Notice that it is necessary to change the class name *constTransport* with *thermoDiffconstTransport* inside the renamed files (this is done with *sed* command).

```
cd $WM_PROJECT_USER_DIR/src/thermophysicalModels/specie/transport
cp -r const thermoDiffconst
cd thermoDiffconst
mv constTransport.C thermoDiffconstTransport.C
mv constTransport.H thermoDiffconstTransport.H
mv constTransportI.H thermoDiffconstTransportI.H
sed -i s/constTransport/thermoDiffconstTransport/g thermoDiffconstTransport*
sed -i s/"const<"/"thermoDiffconst"/g thermoDiffconstTransport.H
cd ../..
cd include
```

Now, we have to include the new code in the thermophysical models types, returning to the *specie* directory folder and access to the *./specie/include* folder where the file *thermoPhysicsTypes.H* is located. Open the file *thermoPhysicsTypes.H* file with the text editor of your preference, and include the *thermoDiffconstTransport.H* header file between *constTransport.H* and *icoPolynomial.H* like in the following example:

```
52 |#include "sutherlandTransport.H"
53 |#include "constTransport.H"
54 |#include "thermoDiffconstTransport.H" //- New Transport Model.
55 |
56 |#include "icoPolynomial.H"
...
```

Then, inside the *namespace Foam* declaration, the following lines are added to create a thermophysics type that contains our new transport model. For *thermo-physics* types based on sensibleEnthalpy:

```
. . .
   62 |namespace Foam
   63 | {
   64
         // thermo physics types based on sensibleEnthalpy
   65 I
           typedef
   66
           thermoDiffconstTransport //- New transport model
   67
            <
   68
              species::thermo
   69
              <
   70
                 hConstThermo
   71 |
                 <
   72
                    perfectGas<specie>
   73
                 >,
   74
                 sensibleEnthalpy
   75
              >
   76
           > thermoDiffconstGasHThermophysics; //- New thermophysical model name
   77
   78 | typedef
   79 | constTransport
```

• • •

As well as the next lines for thermo-physics types based on sensibleInternalEnergy:

```
. . .
   202 // thermo physics types based on sensibleInternalEnergy
   203 typedef
   204 | thermoDiffconstTransport //- New transport model
   205 <
   206
           species::thermo
   207
           <
   208
              eConstThermo
   209
              <
   210
                 perfectGas<specie>
   211
              >,
   212
              sensibleInternalEnergy
   213
   214 > thermoDiffconstGasEThermoPhysics; //- New thermophysical model name
   215
   216 typedef
```

217 | constTransport

• • •

It is important to notice that each line of the new code has to correspond with the marked linenumbers in the file (this can be compared with the accompanying files). Now, return to the folder thermoDiffconst (cd ../transport/thermoDiffconst/) and open the file thermoDiffconstTransport.H. At the thermoDiffconstTransport class declaration add the following lines that correspond to D_{ab} and K_T as new private data members:

```
...
87 | // Private data
88 |
89 | //- Mass diffusivity [m^2/s]
90 | scalar Dab_; //- New data member Dab
91 |
92 | //- Thermodiffusion ratio []
93 | scalar KT_; //- New data member KT
94 |
95 | //- Constant dynamic viscosity [Pa.s]
...
```

And at the *private member function* declaration add the new data members D_{ab} and KT inside the constructor declaration:

```
. . .
   103 //- Construct from components
   104 | inline thermoDiffconstTransport
   105| (
   106
             const Thermo& t,
   107
             const scalar Dab, //- New data member Dab
             const scalar KT, //- New data member KT
   108
   109
             const scalar mu,
   110
             const scalar Pr
   111|);
. . .
```

Finally, at the *member functions section*, add the next function declarations (between *typename()* and dynamic viscosity functions):

```
...
137| return "thermoDiffconst<" + Thermo::typeName() + '>';
138| }
139|
140| //- Mass diffusivity [m^2/s]
141| inline scalar Dab(const scalar p, const scalar T) const; //- New data member Dab
142|
143| //- thermodiffusion coefficient [m^2/s]
144| inline scalar DT(const scalar p, const scalar T) const; //- New data member DT
145|
146| //- Dynamic viscosity [kg/ms]
...
```

Save the file and then open the thermoDiffconstI.H file where the inline functions of the class thermoDiffconstTransport are defined. Write the next code lines at the specified line numbers in the constructors declaration section:

```
. . .
   31 | template<class Thermo>
   32 | inline Foam::thermoDiffconstTransport<Thermo>::thermoDiffconstTransport
   33 (
   34 | const Thermo& t,
           const scalar Dab, //- New data member Dab
   35
           const scalar KT, //- New data member KT
   36
   37
           const scalar mu,
   38
           const scalar Pr
   39 )
   40 | :
   41
           Thermo(t),
   42
           Dab_(Dab), //- New data member Dab
   43
           KT_{(KT)},
                     //- New data member KT
   44
           mu_(mu),
   45
           rPr_(1.0/Pr)
   46 | \{\}
. . .
   49 | template<class Thermo>
   50 | inline Foam::thermoDiffconstTransport<Thermo>::thermoDiffconstTransport
   51 (
   52 | const word& name,
   53
           const thermoDiffconstTransport& ct
   54 | )
   55 :
   56
           Thermo(name, ct),
   57
           Dab_(ct.Dab_), //- New data member Dab
                        //- New data member KT
           KT_(ct.KT_),
   58
   59
           mu_(ct.mu_),
   60
           rPr_(ct.rPr_)
   61 | \{\}
. . .
```

After this addition, two inline functions have to be defined in the member functions section, the first one returns the constant value of D_{ab} (specified as a dictionary entry) and the second one returns the value of D_T calculated as a function of D_{ab} , K_T and T, this addition is done at the beginning of the *member functions* section:

```
. . .
```

```
83 // * * * * *
                        * * * * * * Member Functions
                                                     * * *
                                                                  * * * * * * //
84
85 | template<class Thermo>
86 | inline Foam::scalar Foam::thermoDiffconstTransport<Thermo>::Dab
87 | (
88
        const scalar p,
89
        const scalar T
90 | ) const
91 | {
        return Dab_; //- Constant mass diffusivity for binary mixtures.
92
93 | }
94
```

```
95 |
96 | template<class Thermo>
97 | inline Foam::scalar Foam::thermoDiffconstTransport<Thermo>::DT
98 | (
99 | const scalar p,
100 | const scalar T
101 | const
102 | {
103 | return (KT_*Dab(p, T))/T; //- Thermodiffusion coefficient.
104 | }
```

Finally at the *member operators* and the *friend operators* sections, the two new data members have to be added as dictionary entries, for each operator function. Add the corresponding code lines for D_{ab} and KT data members, at the specified line numbers:

```
149 | Dab_ = ct.Dab_; //- New data member Da
   150 | KT_ = ct.KT_; //- New data member KT
   151 | mu_ = ct.mu_;
   152| rPr_ = ct.rPr_;
   171 | Dab_ = Y1*Dab_ + Y2*st.Dab_; //- New data member Dab
   172| KT_ = Y1*KT_ + Y2*st.KT_;
                                     //- New data member KT
   173 mu_ = Y1*mu_ + Y2*st.mu_;
   174| rPr_ = 1.0/(Y1/rPr_ + Y2/st.rPr_);
. . .
   207| t,
   208 0,
   209 ct1.Dab_, //- New data member Dab
   210 | ct1.KT_, //- New data member KT
   211 ct1.rPr_
. . .
   236| t,
   237 | Y1*ct1.Dab_ + Y2*ct2.Dab_, //- New data member Dab
   238 | Y1*ct1.KT_ + Y2*ct2.KT_,
                                     //- New data member KT
   239 Y1*ct1.mu_ + Y2*ct2.mu_,
   240| 1.0/(Y1/ct1.rPr_ + Y2/ct2.rPr_)
. . .
   240 s*static_cast<const Thermo&>(ct),
   241 | ct.Dab_, //- New data member Dab
   242 ct.KT_,
                   //- New data member KT
   243 | ct.mu_,
. . .
```

. . .

Save the file and close it. For the final step, open the file thermoDiffconstTransport. C and add the next lines as part of the dictionary reading constructor and the dictionary reading member function declaration respectively:

```
...
37 | Thermo(dict),
38 | Dab_(dict.subDict("transport").get<scalar>("Dab")), //- New data member Dab
39 | KT_(dict.subDict("transport").get<scalar>("KT")), //- New data member KT
40 | mu_(dict.subDict("transport").get<scalar>("mu")),
...
56 | os.beginBlock("transport");
```

```
57 | os.writeEntry("Dab", Dab_); //- New data member Dab
58 | os.writeEntry("KT", KT_); //- New data member KT
59 | os.writeEntry("mu", mu_);
```

Save the file and close it. Now the *specie* library can be recompiled. Return to the main folder *thermophysicalModels* using the terminal and write the following commands:

```
cd $WM_PROJECT_USER_DIR/src/thermophysicalModels/
wclean lib specie
wmake libso specie
```

2.2.1 Creating the new thermophysical model: thermoDiffRhoThermo

In order to execute the functions defined in the past section with the *thermophysicalProperties* dictionary, it is necessary to declare and define the new data and function members in a base thermophysical model. At the *basic* directory, one can find a folder called *rhoThermo*, this model constructs a basic thermophysical model based on density, hence it can be used for compressible flow modelling which can be used for binary gas mixture separation modelling.

The first step is to copy the *rhoThermo* folder and rename it as *thermoDiffrhoThermo*, then the liquidThermo.H and liquidThermo.C will be removed as those files correspond to a liquid properties selector function that is not necessary for our implementation.

```
cd $WM_PROJECT_USER_DIR/src/thermophysicalModels/basic/
cp -r rhoThermo thermoDiffrhoThermo
cd thermoDiffrhoThermo
rm liquidThermo*
mv rhoThermo.H thermoDiffrhoThermo.H
mv rhoThermo.C thermoDiffrhoThermo.C
mv rhoThermos.C thermoDiffrhoThermos.C
mv heRhoThermo.H thermoDiffheRhoThermo.H
mv heRhoThermo.C thermoDiffheRhoThermo.C
sed -i s/rhoThermo/thermoDiffrhoThermo/g thermoDiffrhoThermo*
sed -i s/rhoThermo/thermoDiffrhoThermo/g thermoDiffheRhoThermo*
sed -i s/heRhoThermo/thermoDiffheRhoThermo/g thermoDiffheRhoThermo*
sed -i s/heRhoThermo/thermoDiffheRhoThermo/g thermoDiffheRhoThermo*
sed -i s/heRhoThermo/thermoDiffheRhoThermo/g thermoDiffheRhoThermo*
```

Now, all the files have been modified to contain the new class names that correspond to the name of the new thermophysical model (e.g. *rhoThermo* changed to *thermoDiffrhoThermo*). We continue with the new additions by opening the first file *thermoDiffrhoThermo.H* and we add the next lines in the *thermoDiffrhoThermo* class declaration at the *protected data* section and the *member functions* for accessing transport variables section, following the corresponding line counter numbers:

. . .

65 | //- Density field [kg/m^3] 66 | // Named 'thermoDiffrhoThermo' to avoid (potential) conflict with solver density 67 | volScalarField rho_; 68 | 69 | //- Mass diffusivity [m^2/s] 70 | volScalarField Dab_; //New data member Dab 71 | 72 | //- Thermodiffusion coefficient [m^2/s*K] 73 | volScalarField DT_; //New data member DT 74 | 75 | //- Compressibility [s^2/m^2]

```
. . .
   199 //- Dynamic viscosity of mixture for patch [kg/m/s]
   200 | virtual tmp<scalarField> mu(const label patchi) const;
   201
   202 //- Mass diffusivity [m^2/s]
   203 | virtual tmp<volScalarField> Dab() const; //- New data member Dab
   204
   205 | //- Mass diffusivity for patch [m^2/s]
   206 | virtual tmp<scalarField> Dab(const label patchi) const; //- New data member Dab
   207
   208 //- Thermodiffusion coefficient [m^2/s*K]
   209| virtual tmp<volScalarField> DT() const; //- New data member DT
   210
   211 //- Thermodiffusion coefficient for patch [m^2/s*K]
   212 | virtual tmp<scalarField> DT(const label patchi) const; //- New data member DT
   213| };
. . .
```

Save and close the file. Now, open the thermoDiffrhoThermo.C file and add the D_{ab} and D_T function object definitions in the constructors section to construct the new transport coefficients as dimensioned scalars, then at the member functions section, add the member function declarations for both D_{ab} and D_T :

```
. . .
   60 | Dab_ //- New data member Dab
   61 | (
   62
           IOobject
   63
           (
   64
              phasePropertyName("thermo:Dab"),
   65
              mesh.time().timeName(),
   66
              mesh,
              IOobject::NO_READ,
   67
   68
              IOobject::NO_WRITE
   69
            ),
   70
            mesh,
   71
            dimensionSet(0, 2, -1, 0, 0)
   72 | ),
   73
   74 | DT_ //- New data member DT
   75 | (
   76
           IOobject
   77
           (
   78
              phasePropertyName("thermo:DT"),
   79
              mesh.time().timeName(),
   80
              mesh.
   81
              IOobject::NO_READ,
   82
              IOobject::NO_WRITE
   83
           ),
   84
           mesh,
   85
           dimensionSet(0, 2, -1, -1, 0)
   86 | ),
. . .
   140| Dab_ //- New data member Dab
   141 (
   142
           IOobject
```

143	3 (
144	phasePropertyName("thermo:Dab").
145	<pre>b mesh.time().timeName().</pre>
146	mesh.
147	/ IOobject::NO BEAD.
148	I IOobject::NO WRITE
149)).
150) mesh.
15:	dimensionSet(0, 2, -1, 0, 0)
152	2)).
153	3
154	l DT //- New data member DT
155	5 (
156	5 IOobject
157	7 (
158	phasePropertyName("thermo:DT"),
159	<pre>mesh.time().timeName(),</pre>
160	mesh,
16:	IOobject::NO_READ,
162	2 IOobject::NO_WRITE
163	3),
164	h mesh,
165	dimensionSet(0, 2, -1, -1, 0)
166	5),
222 222 222 224 225 226 227 228 230 231 232 233 234 235 236 237 238	<pre>1 (2 IOobject 3 (4 phasePropertyName("thermo:Dab"), 5 mesh.time().timeName(), 5 mesh, 7 IOobject::NO_READ, 8 IOobject::NO_WRITE 9), 9 mesh, 1 dimensionSet(0, 2, -1, 0, 0) 2), 3 4 DT_ //- New data member DT 5 (8 IOobject 7 (9 phasePropertyName("thermo:DT"),</pre>
239	<pre>mesh.time().timeName(),</pre>
240) mesh,
242	Ill IOobject::NO_READ,
242	2 IOobject::NO_WRITE
243	3),
244	h mesh,
245	dimensionSet(0, 2, -1, -1, 0)
246	3),

344 | Foam::tmp<Foam::volScalarField> Foam::thermoDiffrhoThermo::Dab() const

345 { 346 return Dab_; //- New member function Dab 347 } 348 349 350 | Foam::tmp<Foam::scalarField> Foam::thermoDiffrhoThermo::Dab(const label patchi) const 351 { return Dab_.boundaryField()[patchi]; //- New member function Dab 352 353 } 354 355 356 | Foam::tmp<Foam::volScalarField> Foam::thermoDiffrhoThermo::DT() const 357 { return DT_; //- New member function DT 358 359 } 360 361 362 | Foam::tmp<Foam::scalarField> Foam::thermoDiffrhoThermo::DT(const label patchi) const 363 { return DT_.boundaryField()[patchi]; //- New member function DT 364 365 }

Now, open the next file thermoDiffheRhoThermo.H and add the following code lines in the class thermoDiffheRhoThermo declaration:

```
...
69 | volScalarField& alpha,
70 | volScalarField& Dab, //- New member function Dab
71 | volScalarField& DT, //- New member function DT
72 | const bool doOldTimes
...
```

. . .

Then, we follow with the thermoDiffheRhoThermo.C where the new transport coefficients are calculated and updated adding the next lines (Make sure that the previous declarations have a comma at the end, e.g. volScalarField& mu,):

```
. . .
   42 | volScalarField& alpha,
   43 | volScalarField& Dab, //- New data member Dab
   44 | volScalarField& DT, //- New data member DT
   45 | const bool doOldTimes
. . .
   60 | alpha.oldTime(),
   61 | Dab.oldTime(), //- New data member Dab
   62 | DT.oldTime(), //- New data member DT
   63 | true
   74 | scalarField& alphaCells = alpha.primitiveFieldRef();
   75 | scalarField& DabCells = Dab.primitiveFieldRef(); //- New data member Dab
   76 | scalarField& DTCells = DT.primitiveFieldRef(); //- New data member DT
   77
   78 | forAll(TCells, celli)
   97 | alphaCells[celli] = mixture_.alphah(pCells[celli], TCells[celli]);
   98 | DabCells[celli] = mixture_.Dab(pCells[celli], TCells[celli]); //- New data member Dab
```

```
99 | DTCells[celli] = mixture_.DT(pCells[celli], TCells[celli]); //- New data member DT
   100 }
. . .
   108| volScalarField::Boundary& alphaBf = alpha.boundaryFieldRef();
   109| volScalarField::Boundary& DabBf = Dab.boundaryFieldRef(); //- New data member Dab
   110 volScalarField::Boundary& DTBf = DT.boundaryFieldRef(); //- New data member DT
   111
   112 | forAll(pBf, patchi)
. . .
   120| fvPatchScalarField& palpha = alphaBf[patchi];
   121| fvPatchScalarField& pDab = DabBf[patchi]; //- New data member Dab
   122| fvPatchScalarField& pDT = DTBf[patchi]; //- New data member DT
   123
   124 if (pT.fixesValue())
   136| palpha[facei] = mixture_.alphah(pp[facei], pT[facei]);
   137 | pDab[facei] = mixture_.Dab(pp[facei], pT[facei]); //- New data member Dab
   138 | pDT[facei] = mixture_.DT(pp[facei], pT[facei]); //- New data member DT
   139 }
. . .
   156| palpha[facei] = mixture_.alphah(pp[facei], pT[facei]);
   157| pDab[facei] = mixture_.Dab(pp[facei], pT[facei]); //- New data member Dab
   158 | pDT[facei] = mixture_.DT(pp[facei], pT[facei]); //- New data member DT
   159 }
. . .
   183 this->alpha_,
   184| this->Dab_, //- New data member Dab
   185 | this->DT_, //- New data member DT
   186| true
                                // Create old time fields
. . .
   209 | this->alpha_,
   210 | this->Dab_, //- New data member Dab
   211| this->DT_, //- New data member DT
   212 true
                                // Create old time fields
. . .
   239 this->alpha_,
   240 | this->Dab_, //- New data member Dab
   241 | this->DT_, //- New data member DT
   242 false
                      // No need to update old times
. . .
```

Save the file and close it. Finally, we proceed to modify the *thermoDiffrhoThermos.C* file, where the new model is constructed. First, all the unnecessary headers are commented as shown in the following example. Then, inside the Foam namespace, two new thermophysical models will be added and all the current models inside the file can be commented or erased. As the comment lines show in the example, each new model are based on constant enthalpy and a constant internal energy respectively, and each one includes the created thermodiffusion transport model, and the thermodiffusion energy model based on density (*thermoDiffrhoThermo* and *thermoDiffheRhoThermo* respectively), the *thermoDiffrhoThermos.C* file should looks as follows:

26 | *-----*/
27 |
28 | #include "thermoDiffrhoThermo.H"
29 | #include "makeThermo.H"
30 |

```
31 | #include "specie.H"
32 | #include "perfectGas.H"
33 | //#include "incompressiblePerfectGas.H"
34 | //#include "Boussinesq.H"
35 | //#include "rhoConst.H"
36 | //#include "perfectFluid.H"
37 | //#include "PengRobinsonGas.H"
38 | //#include "adiabaticPerfectFluid.H"
39
40 | #include "hConstThermo.H"
41 | #include "eConstThermo.H"
42 | //#include "janafThermo.H"
43 | #include "sensibleEnthalpy.H"
44 | #include "sensibleInternalEnergy.H"
45 | #include "thermo.H"
46
47 | //#include "constTransport.H"
48 | //#include "sutherlandTransport.H"
49 | //#include "WLFTransport.H"
50 | #include "thermoDiffconstTransport.H" //- New transport model for thermodiffusion
51
52 | //#include "icoPolynomial.H"
53 | //#include "hPolynomialThermo.H"
54 | //#include "polynomialTransport.H"
55
56 | #include "thermoDiffheRhoThermo.H"
57 | #include "pureMixture.H"
58
60
61 | namespace Foam
62 | {
63
64 | /* * * * * * * * * * * * * * * private static data * * * * * * * * * * * * * * * /
65
66 | makeThermos
67 (
68thermoDiffrhoThermo,69thermoDiffheRhoThermo,
                              //- Thermophysical model with thermodiffusion
70 | pureMixture,
71 | thermoDiffconstTransport, //- Constant transport model with thermodiffusion
72 | sensibleEnthalpy,
73 | hConstThermo,
74 | perfectGas,
75
       specie
76 | );
77
78 | makeThermos
79 | (
80 | thermoDiffrhoThermo,
                                //- Thermophysical model with thermodiffusion
81 | thermoDiffheRhoThermo,
82 | pureMixture,
83 | thermoDiffconstTransport, //- Constant transport model with thermodiffusion
84 | sensibleInternalEnergy,
```

```
85
     eConstThermo,
86
     perfectGas,
87
     specie
88 | );
89 |
90 | // * * * * * * *
                   * * * * *
                         * * * * * * *
                                   * * *
                                         * * * * * * * * * //
91
92 | } // End namespace Foam
93
```

• • •

Save this file and then proceed with the recompilation of *basic* library, adding the new thermophysical model *thermoDiffheRhoThermo* to the *files* which is the file that contains the main . C source files list of the library. Open the file $$WM_PROJECT_USER_DIR/src/thermophysicalModels/basic/Make/files$ with the text editor of your choice and make sure that it looks like in the following example:

```
basicThermo/basicThermo.C
fluidThermo/fluidThermo.C
psiThermo/psiThermo.C
psiThermo/psiThermos.C
rhoThermo/rhoThermo.C
rhoThermo/liquidThermo.C
rhoThermo/liquidThermo.C
thermoDiffrhoThermo/thermoDiffrhoThermo.C
thermoDiffrhoThermo/thermoDiffrhoThermos.C
derivedFvPatchFields/fixedEnergy/fixedEnergyFvPatchScalarField.C
derivedFvPatchFields/gradientEnergy/gradientEnergyFvPatchScalarField.C
derivedFvPatchFields/mixedEnergy/mixedEnergyFvPatchScalarField.C
derivedFvPatchFields/mixedEnergy/mixedEnergyFvPatchScalarField.C
```

LIB = \$(FOAM_USER_LIBBIN)/libfluidThermophysicalModels

• • •

Finally, at the terminal window, write the following command lines to recompile the new *basic* library.

cd \$WM_PROJECT_USER_DIR/src/thermophysicalModels wclean lib basic wmake libso basic

Chapter 3

Creating a new compressible solver for binary gas-mixtures

The implementation of the species transport equation for binary gas mixtures will be made into one of the current OpenFOAM's compressible solvers, which are one of the set of solvers that allows the utilization of the *thermophysicalProperties* dictionary. The chosen solver is the *rhoSimpleFoam* solver, which is a steady state compressible solver that is used for solving fluid problems with variable density.

3.1 Copying the compressible solver: thermoDiffRhoSimple-Foam

The first thing to do, is avoiding compilation errors that can cause important issues to OpenFOAM's correct performance. One of the most common errors happens when the compilation paths are duplicated or wrongly targeted, causing bugs and interference with OpenFOAM's main code structure. The usual way to do this is to copy the solver code files that will work as the base code for the new implementations, then the main .C file should be renamed, making the necessary changes inside this file and the *Make* folder files, where not only the name of the new solver should be added, the executable address should be changed as well.

In a terminal window, set the OpenFOAM environment (e.g. by typing of-v1906) and add the next lines in order to set the base code files for the new solver.

```
mkdir -p $WM_PROJECT_USER_DIR/applications/solvers/compressible/thermoDiffRhoSimpleFoam
cd $FOAM_SOLVERS/compressible/rhoSimpleFoam
cp -r . $WM_PROJECT_USER_DIR/applications/solvers/compressible/thermoDiffRhoSimpleFoam
cd $WM_PROJECT_USER_DIR/applications/solvers/compressible/thermoDiffRhoSimpleFoam
rm -r overRhoSimpleFoam
rm -r rhoPorousSimpleFoam
mv rhoSimpleFoam.C thermoDiffRhoSimpleFoam.C
sed -i s/"rhoSimpleFoam"/"thermoDiffRhoSimpleFoam"/g thermoDiffRhoSimpleFoam.C
sed -i s/"rhoSimpleFoam"/"thermoDiffRhoSimpleFoam"/g Make/files
sed -i s/"FOAM_APPBIN"/"FOAM_USER_APPBIN"/g Make/files
wclean
wmake
```

Then, check for any error messages at the compilation output in the terminal window, After this, it is a good practice to check that the new solver works with one of the available tutorial cases, this can be done by following the next lines.

```
cd $WM_PROJECT_USER_DIR/run
cp -r $FOAM_TUTORIALS/compressible/rhoSimpleFoam/squareBendLiq ./thermoDiffSquareBendLiq
cd thermoDiffSquareBendLiq
sed -i s/"rhoSimpleFoam"/"thermoDiffRhoSimpleFoam"/g system/controlDict
blockMesh
thermoDiffRhoSimpleFoam >& log &
```

Finally, we can have a look at the log file just to see if the new solver is running properly.

3.2 The CaEqn.H file

The species transport equation will be written in terms of the mass fraction of the reference component C_a , considering a variable density, this equation can be represented as follows,

$$\frac{\partial(\rho C_a)}{\partial t} + \nabla \cdot (\Phi C_a) = \nabla \cdot (\rho D_{ab} \nabla C_a + \rho D_T C_{a0} (1 - C_{a0}) \nabla T)$$
(3.1)

Where Φ represents the total mass flux of the mixture ($\Phi = \rho \vec{v}$). The expression can be rearranged to let all the terms that depend on C_a at the left hand side of the equation so that the thermodiffusion term can be solved explicitly by the solver.

$$\frac{\partial(\rho C_a)}{\partial t} + \nabla \cdot (\Phi C_a) - \nabla \cdot (\rho D_{ab} \nabla C_a) = \nabla \cdot (\rho D_T C_{a0} (1 - C_{a0}) \nabla T)$$
(3.2)

Since the chosen solver which will be used for the new implementation is a steady state solver, the time derivative term will be removed from the last equation, so that, the expression to be used for the new solver looks like the following one,

$$\nabla \cdot (\Phi C_a) - \nabla \cdot (\rho D_{ab} \nabla C_a) = \nabla \cdot (\rho D_T C_{a0} (1 - C_{a0}) \nabla T)$$
(3.3)

Now, continue by opening the text editor of your chice, create a new file called CaEqn.H and add the following lines,

```
//Species transport equation in terms of the reference component 'Ca' mass fraction.
fvScalarMatrix CaEqn
(
    fvm::div(phi, Ca)
    - fvm::laplacian(thermo.Dab()*rho, Ca)
);
solve(CaEqn == fvc::laplacian(thermo.DT()*CaO*(1-CaO)*rho, thermo.T()));
CaEqn.relax();
...
```

Save this file inside the same folder that contains the *thermoDiffRhoSimpleFoam* solver.

3.3 The createFields.H file

The first thing to do is to modify the first lines of the createFields.H file to make that the new solver can work with the new thermophysical model. With the text editor of your preference, open the createFields.H file and then change the text line fluidThermo by thermoDiffrhoThermo, it should looks like in the next example:

```
...
Info<< "Reading thermophysical properties\n" << endl;
autoPtr<thermoDiffrhoThermo> pThermo
(
        thermoDiffrhoThermo::New(mesh)
);
thermoDiffrhoThermo& thermo = pThermo();
thermo.validate(args.executable(), "h", "e");
...
```

The mass fraction of the two components are created inside the createFields.H file, this is done by adding the following lines after the definition of the velocity field, notice that the second component is defined depending on the value of the reference component a.

```
. . .
    //Reference component mass fraction
    volScalarField Ca
    (
    IOobject
    (
    "Ca",
    runTime.timeName(),
    mesh.
    IOobject::READ_IF_PRESENT,
    IOobject::AUTO_WRITE
    ),
    mesh
    );
    //Second component mass fraction
    volScalarField Cb
    (
    IOobject
    (
        "Cb",
        runTime.timeName(),
        mesh,
        IOobject::NO_READ,
        IOobject::AUTO_WRITE
    ),
    1-Ca
    );
. . .
```

The initial mass fraction of the reference component can be defined as a constant entry in a customized dictionary called *initialMassFraction*, the value of the initial mass fraction can be easily changed by creating a dictionary file called (*initialMassFraction*) inside the *constant* folder of any simulation case that works with the thermoDiffRhoSimpleFoam solver. For this purpose, add the next code lines after the turbulence model definition, in the createFields. H file:

```
Info<< "Reading initialMassFraction of Ca\n" << endl;</pre>
   IOdictionary initialMassFraction
    (
       IOobject
       (
         "initialMassFraction",
          runTime.constant(),
          mesh,
          IOobject::MUST_READ,
          IOobject::NO_WRITE
       )
   );
   Info<< "Reading initial mass fraction of component 'a'\n" << endl;
    dimensionedScalar Ca0
    (
       "CaO", dimensionSet(0, 0, 0, 0, 0, 0, 0, 0), initialMassFraction
   );
. . .
```

We finally save the createFields.H file with the new implementations and proceed with the next modifications.

3.4 Final implementations in thermoDiffusionFoam.C file and solver compilation

In this step, the thermoDiffusionFoam.C file should be opened. The header file of the new thermophysical model is included at the begining of the file replacing the "fluid Thermo.H" header, and the CaEqn.H file has to be added before the end of the time-loop, as solution values of temperature and velocity will be needed for the species transport equation calculations. the boundary conditions are corrected as well as the thermophysical variables.

```
U.correctBoundaryConditions();
thermo.T().correctBoundaryConditions();
p.correctBoundaryConditions();
thermo.correct();
#include "CaEqn.H" //Include the species transport equation
Cb = 1 - Ca; //Calculate the mass fraction of the second component b
Ca.correctBoundaryConditions(); //Correct boundary conditions after calculations
//End of new additions
runTime.write();
runTime.printExecutionTime(Info);
```

. . .

Now, let's open the file *Make/options* and replace the link path to the basic library with the user directory path (Change the line: $-I\$(LIB_SRC)/thermophysicalModels/basic/lnInclude \ with -I\$(WM_PROJECT_USER_DIR)/src/thermophysicalModels/basic/lnInclude \)$, then, after the line "<u>EXE_LIBS</u> = \" add the link path to the user's library "-*L*\$(FOAM_USER_LIBBIN) \". Make sure that the options file looks like in the following example:

```
EXE_INC = \setminus
-I$(LIB_SRC)/finiteVolume/cfdTools \
-I$(LIB_SRC)/finiteVolume/lnInclude \
-I$(LIB_SRC)/meshTools/lnInclude \
-I$(LIB_SRC)/sampling/lnInclude \
-I$(LIB_SRC)/transportModels/compressible/lnInclude \
-I$(WM_PROJECT_USER_DIR)/src/thermophysicalModels/basic/lnInclude \
-I$(LIB_SRC)/TurbulenceModels/turbulenceModels/lnInclude \
-I$(LIB_SRC)/TurbulenceModels/compressible/lnInclude \
EXE LIBS = \setminus
-L$(FOAM_USER_LIBBIN) \
-lfiniteVolume \setminus
-lfvOptions \
-lmeshTools \
-lsampling \
-lcompressibleTransportModels \
-lturbulenceModels \
-lcompressibleTurbulenceModels \
-latmosphericModels
```

Save the file and proceed with the compilation of the new solver by entering the following command lines in the terminal window.

cd \$WM_PROJECT_USER_DIR/applications/solvers/compressible wclean thermoDiffRhoSimpleFoam wmake thermoDiffRhoSimpleFoam

Chapter 4 Tutorial set up

To test the new implementations made in the *thermophysicalModels* library, a basic test case will be set up. The case consist on a simple 2-D simulation of a *thermodiffusion* cell with a 50% - 50% mass fraction mixture of He and N_2 entering to the domain at $4 \cdot 10^{-06} \frac{kg}{s}$. The domain resembles a parallel plate flow system where a temperature gradient is induced by setting the higher temperature in the upper wall and the lower temperature in the bottom wall. Table 4.1 shows the thermophysical constant values that are necessary as initial conditions for this test case.



Figure 4.1: Sketch of the thermodiffusion cell, 2-D geometry.

Since the purpose of this tutorial is to show how the *thermophysicalModels* library works with the new defined mass transport coefficients in a compressible solver, only the *thermophysicalProperties* and the custom *initialMassFraction* dictionaries are explained here, the rest of the case files are compelled in the *Appendix* section.

Description	Symbol	Value and units
Mass fraction of He	C_a	0.5
Mass fraction of N_2	C_b	0.5
Mixture mole weight	М	$7.00125[\frac{kg}{mol}]$
Mixture dynamic viscosity	μ	$1.969 \cdot 10^{-5} [\frac{kgm}{s}]$
Mass diffusivity	D_{ab}	$1.969 \cdot 10^{-5} [\frac{m^2}{s}]$
Entalpy of formation	h_f	$0[\frac{kgm^2}{s^2mol}]$
Mixture specific heat	c_p	$3112.452[\frac{m^2}{s^2K}]$
Prandtl	Pr	0.45
Thermodiffusion ratio	K_T	20.33

Table 4.1: Mixture therrmophysical properties

4.1 The thermophysical properties and initial MassFraction dictionaries

In the *thermoPhysicalProperties* dictionary, one has to specify the entries for the new thermophysical and transport models, the reader can compare the values given in table 4.1 with the entry values of the following example of the *thermophysicalProperties* dictionary.

```
FoamFile
ſ
    version
               2.0;
    format
               ascii;
               dictionary;
    class
    location
                "constant";
    object
                thermophysicalProperties;
}
// * * * * * * * *
                         *
                           thermoType
{
                   thermoDiffheRhoThermo; //- New thermophysical model
    type
    mixture
                  pureMixture;
    transport
                   thermoDiffconst; //- New transport model
                   hConst;
    thermo
    equationOfState perfectGas;
    specie
                   specie;
    energy
                   sensibleEnthalpy;
}
mixture
{
specie
ł
             7.00125;
  molWeight
}
thermodynamics
{
              3112.452;
   Ср
  Hf
              0;
}
transport
{
              1.969e-05;
  mu
  \Pr
             0.45;
  Dab
              7.5e-5;
  KΤ
              20.3;
}
}
```

As mentioned in the previous chapter, the initial mass fraction of the reference component can be specified in a user-defined dictionary, called *initialMassFraction*, thereby, that parameter can be easily changed without recurring to solver modifications.

```
FoamFile
{
   version
                2.0;
                ascii;
   format
   class
                dictionary;
                "constant";
   location
   object
                initialMassFraction:
}
// *
//Initial mass
                fraction of component "a" N2 : mixture N2-He
Ca0
                      0.5:
```

4.2 Simulation results post-processing

Since the initial mass fraction of the reference component and the maximum and minimum values of temperature are known, the equation (1.6) can be easily integrated to get the final mass fraction value of the reference component. The integrated equation is given below,

$$C_a - C_{a0} = -K_T C_{a0} (1 - C_{a0}) \ln \left[\frac{T_H}{T_C}\right]$$
(4.1)

By substituting the values given in table 4.1 the mass fraction value of the reference component is $C_{\{a,calc\}} = 0.5328$, this value can be compared with the maximum value of the component a curve, shown in the figure 4.2 which is approximately $C_{\{a,sim\}} = 0.54$. This gives a relative error value of approximately 1.35%.

Therefore, not only the new implementation works fine with the solver, but also it is possible to get accurate results for a simple steady state simulation case.



Figure 4.2: Mass fraction variation along the domain center line in the X direction, for both mixture components $(He - N_2)$

With the help of figures 4.3 and 4.4, it is easy to see how the concentration variation depends on the temperature field. While the temperature reaches a uniform gradient value along Y axis, the mass fraction of the reference component settles on an averaged value along the same direction, thus, according to equation 4.1, in a steady state flow, the concentration of species depends on the averaged temperature value and it is independent of the gradient direction. Finally, one last observation is that in accordance to the theory [4], when the value of K_T is positive, the light component will move towards the hotter region and the heavy component will move towards the coldest region, causing a spatial concentration variation of the mixture depending on the local temperature values (the contrary effect takes place when the value of K_T is negative). The reader can run his own test case by copying all the case files given in the Appendix section (Or preferably, by downloading the complementary files that come with this document).



Figure 4.5: Contour field of N_2 Mass fraction.

The method shown in this report can be extended for further additions to the *thermophysicalModels* library, such as new transport models that include calculation models for a variable mass diffusivity and a variable thermodiffusion coefficient. Further coupling between these thermodiffusion transport models, and other thermodynamic and mixture models can be done to assure stronger calculations that can deliver more accurate results. The interested reader is invited to test this new implementations for different fluid mixture types and for different thermodiffusion problems, if there is an interest in extend this work for more complex problems, enquires can be made to the author via email (pmjlab@leeds.ac.uk).

Appendix: Simulation case files

system folder

controlDict file FoamFile ſ version 2.0; format ascii; class dictionary; location "system"; object controlDict; } application thermoDiffrhoSimpleFoam; startFrom latestTime: startFrom latestTime; 0; startTime stopAt endTime; 0.4; 1e-04; //endTime //deltaT writeControl adjustableRunTime; //writeInterval 2e-3; 0; purgeWrite writeFormat ascii; //binary; writePrecision 6; writeCompression off; timeFormat general; timePrecision 6; runTimeModifiable yes; adjustTimeStep yes; maxCo 1; // SIMPLE LOOP endTime 2000; deltaT 1; writeInterval 10;

blockMesh file

```
FoamFile
{
   version
            2.0;
   format
            ascii;
   class
           dictionary;
   object
           blockMeshDict;
7
convertToMeters 1;
vertices
(
   (-0.05 - 0.005 0)
   (0.05 -0.005 0)
   (0.05 \ 0.005 \ 0)
   (-0.05 \ 0.005 \ 0)
   (-0.05 -0.005 0.005)
```

```
(0.05 - 0.005 0.005)
    (0.05 0.005 0.005)
    (-0.05 0.005 0.005)
);
blocks
(
    hex (0 1 2 3 4 5 6 7) (400 40 1) simpleGrading (
    1
    (
        (0.2 0.35 4) // 20% y-dir, 35% cells, expansion = 4
(0.6 0.3 1) // 60% y-dir, 30% cells, expansion = 1
(0.2 0.35 0.25) // 20% y-dir, 35% cells, expansion = 0.25 (1/4)
    )
    1
    )
);
edges
(
);
boundary
(
topWall
{
     type wall;
     faces
     (
        (3762)
     );
}
bottomWall
{
     type wall;
     faces
     (
     (0 1 5 4)
     );
}
inlet
{
     type patch;
     faces
     (
         (0 4 7 3)
     );
}
outlet
{
     type patch;
     faces
     (
       (2651)
     );
}
frontAndBack
{
     type empty;
     faces
     (
         (0 3 2 1)
         (4567)
     );
}
);
mergePatchPairs
(
);
// ****
```

fvSchemes file

```
FoamFile
ł
    version
               2.0;
    format
               ascii;
               dictionary;
    class
    location
               "system";
    object
               fvSchemes;
}
ddtSchemes
{
    default
                   Euler;
}
gradS chemes
ſ
                   leastSquares;
    default
    grad(U)
                   fourth ;
}
divSchemes
{
    default
                   none;
    div(phi,U)
                   Gauss limitedLinearV 1;
    div(phi,e)
                   Gauss limitedLinear 1;
                  Gauss limitedLinear 1;
    div(phi,K)
    div(phiv,p)
                   Gauss limitedLinear 1;
                   Gauss limitedLinear 1;
    div(phi,Ca)
    div(meshPhi,p) Gauss limitedLinear 1;
    div(phi,h)
                   Gauss limitedLinear 1;
    div(phi,Ekp)
                   Gauss limitedLinear 1;
    div(((rho*nuEff)*dev2(T(grad(U))))) Gauss linear;
}
laplacianSchemes
{
    default
                                Gauss linear corrected;
}
interpolationSchemes
{
               limitedLinear phi 1;
    default
}
snGradSchemes
ſ
    default
               orthogonal;
}
```

fvSolution file

```
FoamFile
{
            2.0;
   version
   format
            ascii;
   class
            dictionary;
   location
            "system";
   object
            fvSolution;
}
solvers
{
"(p|rho|Ca)"
{
   solver
               GAMG;
  tolerance
               1e-06;
   relTol
              0.05;
   smoother
               symGaussSeidel;
   nCellsInCoarsestLevel 200;
}
U
```

```
{
                    smoothSolver;
    solver
    smoother
                    symGaussSeidel;
    nSweeps
                    4;
    tolerance
                    1e-05;
    relTol
                    0.1;
    minIter
                    1;
}
h
{
                    PBiCGStab;
    solver
    preconditioner DILU;
   nSweeps
                    2;
    tolerance
                    1e-05;
    relTol
                    0.1;
    minIter
                    1;
}
"(p|rho|Ca)Final"
{
    $p;
                    1e-05;
    tolerance
   relTol
                    0;
    minIter
                    1;
}
UFinal
{
    $U;
                    1e-05;
    tolerance
    relTol
                    0.1;
    minIter
                    1;
}
hFinal
{
    $e;
    tolerance
                    1e-05;
    relTol
                    0;
    minIter
                    1;
}
}
SIMPLE
{
    nNonOrthogonalCorrectors 4;
    rhoMin
                   0.3;
    rhoMax
                    1.4;
                    90000;
    pMin
    pMax
                    110000;
    transonic
                    false;
    pRefCell = 0;
    pRefValue = 1e5;
    //consistent
                      yes;
    residualControl
    {
                       1e-3;\\
        р
        U
                        1e-4;\\
                       1e-3;
        е
                       1e-3;
        Ca
        // possibly check turbulence fields
        // "(k|epsilon|omega)" 1e-3;
    }
    // 2.4.x
    rhoMin rhoMin [1 -3 0 0 0 0 0] 0.3;
     rhoMax rhoMax [1 -3 0 0 0 0 0] 1.4;
}
relaxationFactors-SIMPLE
{
     fields
     {
                     0.3;
     р
```

```
rho
                   0.05;
    }
    equations
    {
    U
                   0.7;
     // "(k|epsilon)" 0.7;
                    0.5;
     е
                    0.5;
    Ca
    ".*Final"
                    1.0;
    }
}
relaxationFactors-PIMPLE
{
    equations
    {
U
                      0.95;
    // "(k|epsilon)" 0.95;
                    0.95;
     е
     Ca
                     0.95;
       ".*Final"
                     1.0;
    }
}
relaxationFactors { $relaxationFactors-SIMPLE }
```

constant folder

thermoPhysicalProperties file

```
FoamFile
{
              2.0;
    version
             ascii;
    format
             dictionary;
    class
    location "constant";
    object
             thermophysicalProperties;
}
thermoType
{
                thermoDiffheRhoThermo;
    type
                pureMixture;
    mixture
    transport
                 thermoDiffconst;
    thermo
                hConst;
    equationOfState perfectGas;
    specie
                 specie;
                 sensibleEnthalpy;
    energy
}
mixture
{
specie
{
   molWeight 7.00125;
}
thermodynamics
{
   Cp
             3112.452;
   Hf
             0;
}
transport
{
             1.969e-05;
   mu
             0.45;
   Pr
   Dab
             7.5e-5;
   ΚT
             20.3;
}
}
```

initialMassFraction file

```
FoamFile
{
          2.0;
   version
   format
          ascii;
           dictionary;
   class
   location
          "constant";
          initialMassFraction;
   object
}
//Initial mass fraction of component "a" N2 : mixture N2-He
Ca0
              0.5;
```

turbulenceProperties

θ folder

Ca file

```
FoamFile
ſ
   version 2.0;
        ascii;
  format
  class
           volScalarField;
  location "0";
           Ya;
  object
}
dimensions [0 0 0 0 0 0];
internalField uniform 0.5;
boundaryField
{
inlet
{
              uniformFixedValue;
   type
   uniformValue
              constant 0.5;
}
outlet
{
               inletOutlet;
   type
   value
              $internalField;
   inletValue
               uniform 0;
}
topWall
{
               zeroGradient;
   type
}
bottomWall
{
               zeroGradient;
  type
}
frontAndBack
{
               empty;
   type
}
```

}

U file FoamFile

```
{
   version
            2.0;
   format
           ascii;
   class
             volVectorField;
   object
             U;
}
[0 1 -1 0 0 0 0];
dimensions
internalField uniform (0 0 0);
boundaryField
{
inlet
{
               flowRateInletVelocity;
   type
  massFlowRate
               constant 4e-06;
               uniform (0 0 0);
  value
}
outlet
{
               inletOutlet;
   type
               $internalField;
   value
               uniform (0 0 0);
   inletValue
}
topWall
{
   type
         noSlip;
}
bottomWall
{
          noSlip;
   type
}
frontAndBack
{
   type
          empty;
}
}
P file
FoamFile
{
            2.0;
   version
   format
            ascii;
   class
             volScalarField;
   object
             p;
}
[1 -1 -2 0 0 0 0];
dimensions
internalField
            uniform 101325;
boundaryField
{
inlet
{
   type
               zeroGradient;
}
outlet
{
               fixedValue;
  type
               $internalField;
   value
}
topWall
{
```

```
type zeroGradient;
}
bottomWall
{
   type zeroGradient;
}
frontAndBack
{
   type empty;
}
}
```

T file

```
FoamFile
{
           2.0;
   version
   format ascii;
   class
            volScalarField;
            "0";
   location
   object
            Τ;
}
[0 0 0 1 0 0 0];
dimensions
internalField uniform 298.15;
boundaryField
{
inlet
{
               uniformFixedValue;
   type
   uniformValue constant 298.15;
}
outlet
{
               inletOutlet;
   type
              $internalField;
  value
  inletValue
              uniform 298.15;
}
topWall
{
   type
              uniformFixedValue;
   uniformValue
             constant 310;
}
bottomWall
{
               uniformFixedValue;
   type
   uniformValue
               constant 298.15;
}
frontAndBack
{
   type
              empty;
}
}
```

Study questions

- 1. How do temperature gradients affect mass diffusivity in binary gas mixtures?
- 2. What is the thermodiffusion ratio? How is it related to mass diffusivity processes?
- 3. Describe briefly the structure of the *thermophysicalModels* library, how are the thermophysical properties of a fluid obtained/calculated when running a simulation case?
- 4. In the *thermophysicalModels* code context, why is it necessary to declare the data members and member functions of a transport model in a base thermophysical model? (it can also be a thermodynamic or state equation model).
- 5. In a solver code where thermophysical models are required, how are thermophysical models declared? Which file contains that declaration?
- 6. Explain briefly the entries of the *thermophysicalProperties* dictionary (this is related with the structure of *thermophysicalModels* library).

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