Turbulence-chemistry interaction in OpenFOAM and how to implement a dynamic PaSR model for LES of turbulent combustion

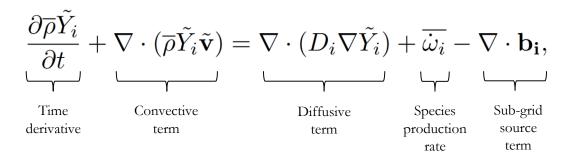
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LES of turbulent combustion

- Large Eddy Simulation (LES) is a method for simulating unsteady turbulent flows
- Large turbulent structures (~80% of kinetic energy) are resolved while smaller structures are modeled
- In reactive flow, a transport equation is solved for the mass fraction \tilde{Y}_i of each species i



The filtering problem

- LES uses $\overline{\dot{\omega}_i}$, the filtered reaction rate of species *i*
- Problem: How to compute $\overline{\dot{\omega}_i}$ if all the complicated subgrid chemistry is unresolved?
- My impression after speaking with colleagues:
 - There is no simple way to compute $\overline{\dot{\omega}_i}$
 - The underlying assumptions of many existing models are very questionable
 - We should all be scared and confused

$\overline{\dot{\omega}_i}$

Some turbulence-chemistry interaction models (simplified)

- In the **Perfectly** Stirred Reactor (PSR) model, each mesh cell is a homogeneous reactor. $\overline{\dot{\omega}_{i,PSR}} = \dot{\omega}_i$
- The Eddy Dissipation Concept (EDC) model assumes that $\overline{\dot{\omega}_i} \approx \gamma^* \overline{\dot{\omega}_{i,PSR}}$. Simply put, turbulence causes reactions to only take place in localized, anisotropic pockets. The volume fraction γ^* of these pockets is determined from turbulence. Note: $0 \le \gamma^* \le 1$.
- The Fractal Model (FM) is similar to EDC, but computes γ^* based on a cascade model.
- (Flamelet models can also be used to get around the $\overline{\dot{\omega}_i}$ problem, but in a different way.)



The Partially Stirred Reactor (PaSR) model (also simplified)

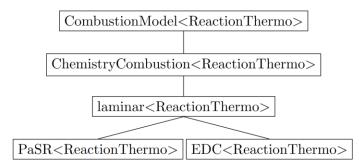
• The Partially Stirred Reactor (PaSR) model also assumes that $\overline{\dot{\omega}_i} \approx \gamma^* \overline{\dot{\omega}_{i,PSR}}$, but attempts to take **both turbulence and chemistry** into account when estimating γ^* :

$$\gamma^* = \frac{\tau_c}{\tau_c + \tau^*}$$

- τ_c is the chemical time scale
- τ^* is the mixing time scale
- Simply put: if turbulent mixing is fast relative to chemistry, reactions take place in a wellmixed environment.

Turbulence-chemistry interaction in OpenFOAM

- The laminar (a.k.a. PSR), EDC and PaSR models are implemented as classes in OpenFOAM.
- These inherit from ChemistryCombustion, which gives them access to the chemistry model



Turbulence-chemistry interaction in OpenFOAM

The model classes contain three crucial functions.

//- Correct combustion rate
virtual void correct();

//- Fuel consumption rate matrix.
virtual tmp<fvScalarMatrix> R(volScalarField& Y) const;

//- Heat release rate [kg/m/s3]
virtual tmp<volScalarField> Qdot() const;

PaSR in OpenFOAM

The PaSR class has the following definitions for R and Qdot.

```
template<class ReactionThermo>
Foam::tmp<Foam::fvScalarMatrix>
Foam::combustionModels::PaSR<ReactionThermo>::R(volScalarField& Y) const
    return kappa_*laminar<ReactionThermo>::R(Y);
}
template<class ReactionThermo>
Foam::tmp<Foam::volScalarField>
Foam::combustionModels::PaSR<ReactionThermo>::Qdot() const
ſ
    return tmp<volScalarField>
        new volScalarField
        (
            this->thermo().phasePropertyName(typeName + ":Qdot"),
            kappa_*laminar<ReactionThermo>::Qdot()
        )
    );
```

PaSR in OpenFOAM

The correct function calculates γ^* (here referred to as kappa_). The crucial step is the following loop.

Computing the time scales

- So, how do we compute τ_c and τ^* ?
- In the code we just saw, $\tau^* = \tau_K = \sqrt{\nu/\varepsilon}$
- According to Sabelnikov & Fureby (2013), $\tau^* = \sqrt{\tau_K \tau_\Delta}$ and $\tau_\Delta = \Delta/\nu'$ (time scale of subgrid stretch). This is a better fit for the anisotropic shape of reactive fine structures.
- τ_c is a more complicated story. A review article from 2020 lists 9 different ways to compute it (Wartha *et al.* (2021)) and I know of at least one more.

Chemical time scale in OpenFOAM

OpenFOAM uses the following expression for τ_c :

$$\tau_{c} = \sum_{j=1}^{n_{R}} \frac{c_{tot}}{\sum_{i=1}^{N_{s,RHS}} v_{i,j} k_{f,j}}$$

In words: "Each reaction contributes the total molar concentration c_{tot} divided by the production rate [kmol/m³/s] of all species on its Right-Hand Side (RHS). The chemical time scale is the sum of all such contributions."

Problems

- Using $\tau^* = \tau_K$ does not take anisotropy into account
- The OpenFOAM method for computing τ_c is not made with PaSR in mind
- Why is only the RHS of the reactions considered? What does this mean for reversible reactions?
- All methods I've seen compute a single τ_c and then use that to scale all reactions equally. This is a bit strange, as reactions can have wildly different time scales
- Using a single τ_c probably makes sense when you only have 1-4 reactions
- Nowadays we use chemical mechanisms with 200-300 reactions though...

A new model

To tackle these problems, I propose a new model: foxPaSR.

- Computes separate chemical time scales for all reactions
- $\tau^* = \sqrt{\tau_K \tau_\Delta}$
- Takes reversible reactions into account
- Suitable for reaction mechanisms of any size
- Does not require additional parameters



The foxPaSR algorithm

In each cell, at each time step:

- 1. Compute $\tau^* = \sqrt{\tau_k \tau_\Delta}$.
- 2. Use the PSR model to compute the rates of all reactions.
- 3. For **each** reaction *j*, first identify if the reverse rate is bigger than the forward rate. If so, the reverse reaction is considered to have its reactants on the RHS.
- 4. For each reaction *j*, identify the reactant species *i* with the shortest residence time, defined as $\tilde{Y}_i \bar{\rho} / v_{r,i} k_r$. This residence time is chosen as τ_c for the reaction.
- 5. For each reaction *j*, compute $\gamma_j^* = \frac{\tau_{c,j}}{\tau_{c,j} + \tau^*}$.
- 6. Compute the filtered reaction rates by summing the contributions of all reactions: $\overline{\dot{\omega}_i} = \sum_{j=1}^{n_R} \gamma_j^* \overline{\dot{\omega}_{i,j,PSR}}$





- foxPaSR is a heavily modified version of the existing PaSR class.
- Production rates are modified inside the correct function, while the R and Qdot functions just return the already modified rates

```
template<class ReactionThermo>
Foam::tmp<Foam::fvScalarMatrix>
Foam::combustionModels::foxPaSR<ReactionThermo>::R(volScalarField& Y) const
{
    return laminar<ReactionThermo>::R(Y);
}
```

foxPaSR has three new members:

- Y_, the species mass fractions
- reactions_, the list of reactions
- modRR_, the locally modified production rates of all species

```
Y_(this->thermo().composition().Y()),
reactions_
(
    dynamic_cast<const reactingMixture<gasHThermoPhysics>&>(this->thermo())
),
modRR_(this->chemistryPtr_->nSpecie())
```



The correct function consists of two loops:

- an outer loop, over the list of reactions
- an inner loop, over all cells

```
forAll(reactions_, ri)
{
    label refSpecies = reactions_[ri].lhs()[0].index;
    const scalar& refW =
        this->thermo().composition().W(refSpecies);
    scalarField refRR =
        this->chemistryPtr_->calculateRR(ri, refSpecies)/refW;
```

```
forAll(epsilon, celli)
{
    // To be filled in...
}
```



For reaction ri, the variable refRR represents the rate at which kmol are transferred from the LHS to the RHS. It can be negative.

In each cell, a residence time for each reactant species is computed.

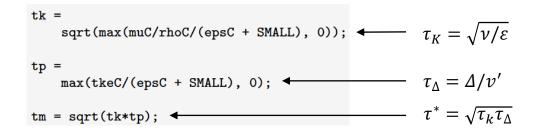
```
tcSpecies =
    mag(rhoC*Y_[i][celli]/(refRRC*W*stoichCoeff));
```

The shortest of these is chosen as the chemical time scale.

```
if (tcSpecies < tc)
{
    tc = tcSpecies;
}</pre>
```



The mixing time scale is also computed.







The reaction-specific volume fraction γ_j^* (kappa) is then computed following the standard PaSR approach. If the mixing is very fast or the chemistry very slow, $\gamma_j^* = 1$ as in the PSR model.

```
if (tm > SMALL && tc < GREAT)
{
    kappa = tc/(tc + tm);
}
else
{
    kappa = 1.0;
}</pre>
```

The scaled production rates in reaction ri are then added to the total modified production rates.

```
modRR_[i][celli] -=
    kappa*refRRC*W*stoichCoeff;
```

When all reactions have been looped over, the production rates initially computed by PSR are replaced with the modified production rates.

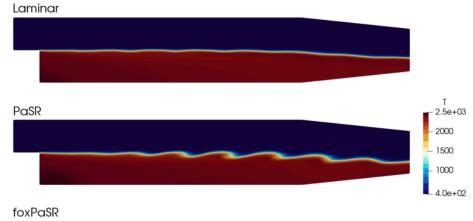
```
forAll(modRR_, i)
{
    this->chemistryPtr_->RR(i) = modRR_[i];
}
```

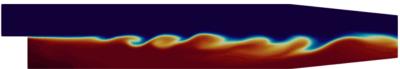
The modified production rates will now be used in the energy and species transport equations.



Some simple results

- Premixed n-heptane flame
- Pitz-Daily geometry with side walls
- Coarse-grained LES with LDKM
- Re $\approx 50\ 000$
- $\phi = 0.75$
- 340 000 hex cells





References

- Sabelnikov, V.; Fureby, C. LES Combustion Modeling for High Re Flames using a Multi-phase Analogy. Combust. Flame **2013**, 160, 83.
- Wartha, E.; Bösenhofer, M.; Harasek, M. Characteristic Chemical Time Scales for Reactive Flow Modeling. Combust. Sci. Technol. **2021**, 193, 2807.

Thanks for listening!



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