

# ThermophysicalModels library in OpenFOAM-2.3.x

*How to implement a new thermophysical model*

Teaching within: CFD with OpenSource software (TME050)

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## Exemples of thermodynamic and transport properties:

- density  $\rho$  → *equation of state (Ia, see slide 4)*
  - heat capacity  $C_v$ ,  $C_p$
  - internal energy  $e$ , enthalpy  $h$  → *thermophysical properties ( Ib, see slide 5)*
  - diffusivity  $D, \alpha$  ...
  - viscosity  $\mu$
  - thermal conductivity  $\kappa$
  - electric conductivity
- *transport properties (Ic, see slide 6)*

## Depend on:

- temperature  $T$
- pressure  $P$  → *Pure mixture (Ia-Ic, see slide 4-6)*
- fluid (possibly solid) composition → *Mixture (II, see slide 7)*

**Involved in** heat transfer, compressible flow, multiphase problems, combustion, etc.

# Content

*Slides*

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- Implement a new transport property ( $\kappa$ ) ..... 31-42
  - Import, declare, define, link to solver, run a case
- Implement a new equation of state (for  $\rho$ )  
and new thermodynamic properties ( $h$  and  $C_p$ ) ..... 43-58
  - Import, declare, define, link to solver, run a case

## Ia Equation of State — equationOfState

---

adiabaticPerfectFluid	Adiabatic perfect gas equation of state
icoPolynomial	Incompressible polynomial equation of state, <i>e.g.</i> for liquids
perfectFluid	Perfect gas equation of state
incompressiblePerfectGas	Incompressible gas equation of state using a constant reference pressure. Density only varies with temperature and composition
rhoConst	Constant density equation of state

## **Ib** Basic thermophysical properties — thermo

---

eConstThermo	Constant specific heat $c_p$ model with evaluation of internal energy $e$ and entropy $s$
hConstThermo	Constant specific heat $c_p$ model with evaluation of enthalpy $h$ and entropy $s$
hPolynomialThermo	$c_p$ evaluated by a function with coefficients from polynomials, from which $h$ , $s$ are evaluated
janafThermo	$c_p$ evaluated by a function with coefficients from JANAF thermodynamic tables, from which $h$ , $s$ are evaluated

**Ic**

## Transport properties — transport

---

constTransport

Constant transport properties

polynomialTransport

Polynomial based temperature-dependent transport properties

sutherlandTransport

Sutherland's formula for temperature-dependent transport properties

## II Mixture properties — mixture

pureMixture	General thermophysical model calculation for passive gas mixtures
homogeneousMixture	Combustion mixture based on normalised fuel mass fraction $b$
inhomogeneousMixture	Combustion mixture based on $b$ and total fuel mass fraction $f_t$
veryInhomogeneousMixture	Combustion mixture based on $b$ , $f_t$ and unburnt fuel mass fraction $f_u$
basicMultiComponentMixture	Basic mixture based on multiple components
multiComponentMixture	Derived mixture based on multiple components
reactingMixture	Combustion mixture using thermodynamics and reaction schemes
egrMixture	Exhaust gas recirculation mixture
singleStepReactingMixture	Single step reacting mixture

→ Ia, Ib & Ic

→ Ia, Ib & Ic for all species & suited mixing rules

## III Thermophysical model — thermoModel **Combines Ia, Ib, Ic & II**

---

hePsiThermo	General thermophysical model calculation based on compressibility $\psi$
heRhoThermo	General thermophysical model calculation based on density $\rho$
psiReactionThermo	Calculates enthalpy for combustion mixture based on $\psi$
psiuReactionThermo	Calculates enthalpy for combustion mixture based on $\psi_u$
rhoReactionThermo	Calculates enthalpy for combustion mixture based on $\rho$
heheupsiReactionThermo	Calculates enthalpy for unburnt gas and combustion mixture

# Models available in OpenFOAM (for a given solver)

(6/8)

**Example:** go to tutorials/heatTransfer/buoyantSimpleFoam/buoyantCavity/constant in thermophysicalProperties change "transport const" to "transport dummy" and run the solver buoyantSimpleFoam; it returns the list of thermophysical models available for this solver:

```
Valid rhoThermo types are:
type III mixture II
Ic Ib Ia
transport thermc equationOfState specie energy
heRhoThermo homogeneousMixture const hConst incompressiblePerfectGas specie sensibleEnthalpy
heRhoThermo homogeneousMixture const hConst perfectGas specie sensibleEnthalpy
heRhoThermo homogeneousMixture sutherland janaf incompressiblePerfectGas specie sensibleEnthalpy
heRhoThermo homogeneousMixture sutherland janaf perfectGas specie sensibleEnthalpy
heRhoThermo inhomogeneousMixture const hConst incompressiblePerfectGas specie sensibleEnthalpy
heRhoThermo inhomogeneousMixture const hConst perfectGas specie sensibleEnthalpy
```

... (+ 2 following slides)

# Models available in OpenFOAM (for a given solver)

(7/8)

heRhoThermo	inhomogeneousMixture	sutherland	janaf	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	inhomogeneousMixture	sutherland	janaf	perfectGas	specie	sensibleEnthalpy
heRhoThermo	multiComponentMixture	const	hConst	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	multiComponentMixture	const	hConst	incompressiblePerfectGas	specie	sensibleInternalEnergy
heRhoThermo	multiComponentMixture	const	hConst	perfectGas	specie	sensibleEnthalpy
heRhoThermo	multiComponentMixture	const	hConst	perfectGas	specie	sensibleInternalEnergy
heRhoThermo	multiComponentMixture	polynomial	hPolynomial	icoPolynomial	specie	sensibleEnthalpy
heRhoThermo	multiComponentMixture	polynomial	hPolynomial	icoPolynomial	specie	sensibleInternalEnergy
heRhoThermo	multiComponentMixture	sutherland	janaf	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	multiComponentMixture	sutherland	janaf	incompressiblePerfectGas	specie	sensibleInternalEnergy
heRhoThermo	multiComponentMixture	sutherland	janaf	perfectGas	specie	sensibleEnthalpy
heRhoThermo	multiComponentMixture	sutherland	janaf	perfectGas	specie	sensibleInternalEnergy
heRhoThermo	pureMixture	const	hConst	adiabaticPerfectFluid	specie	sensibleEnthalpy
heRhoThermo	pureMixture	const	hConst	adiabaticPerfectFluid	specie	sensibleInternalEnergy
heRhoThermo	pureMixture	const	hConst	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	pureMixture	const	hConst	incompressiblePerfectGas	specie	sensibleInternalEnergy
heRhoThermo	pureMixture	const	hConst	perfectFluid	specie	sensibleEnthalpy
heRhoThermo	pureMixture	const	hConst	perfectFluid	specie	sensibleInternalEnergy
heRhoThermo	pureMixture	const	hConst	perfectGas	specie	sensibleEnthalpy
heRhoThermo	pureMixture	const	hConst	perfectGas	specie	sensibleInternalEnergy
heRhoThermo	pureMixture	const	hConst	rhoConst	specie	sensibleEnthalpy
heRhoThermo	pureMixture	const	hConst	rhoConst	specie	sensibleInternalEnergy
heRhoThermo	pureMixture	kineticAr	hKineticAr	rhoKineticAr	specie	sensibleEnthalpy
heRhoThermo	pureMixture	polynomial	hPolynomial	PengRobinsonGas	specie	sensibleEnthalpy
heRhoThermo	pureMixture	polynomial	hPolynomial	icoPolynomial	specie	sensibleEnthalpy
heRhoThermo	pureMixture	polynomial	hPolynomial	icoPolynomial	specie	sensibleInternalEnergy

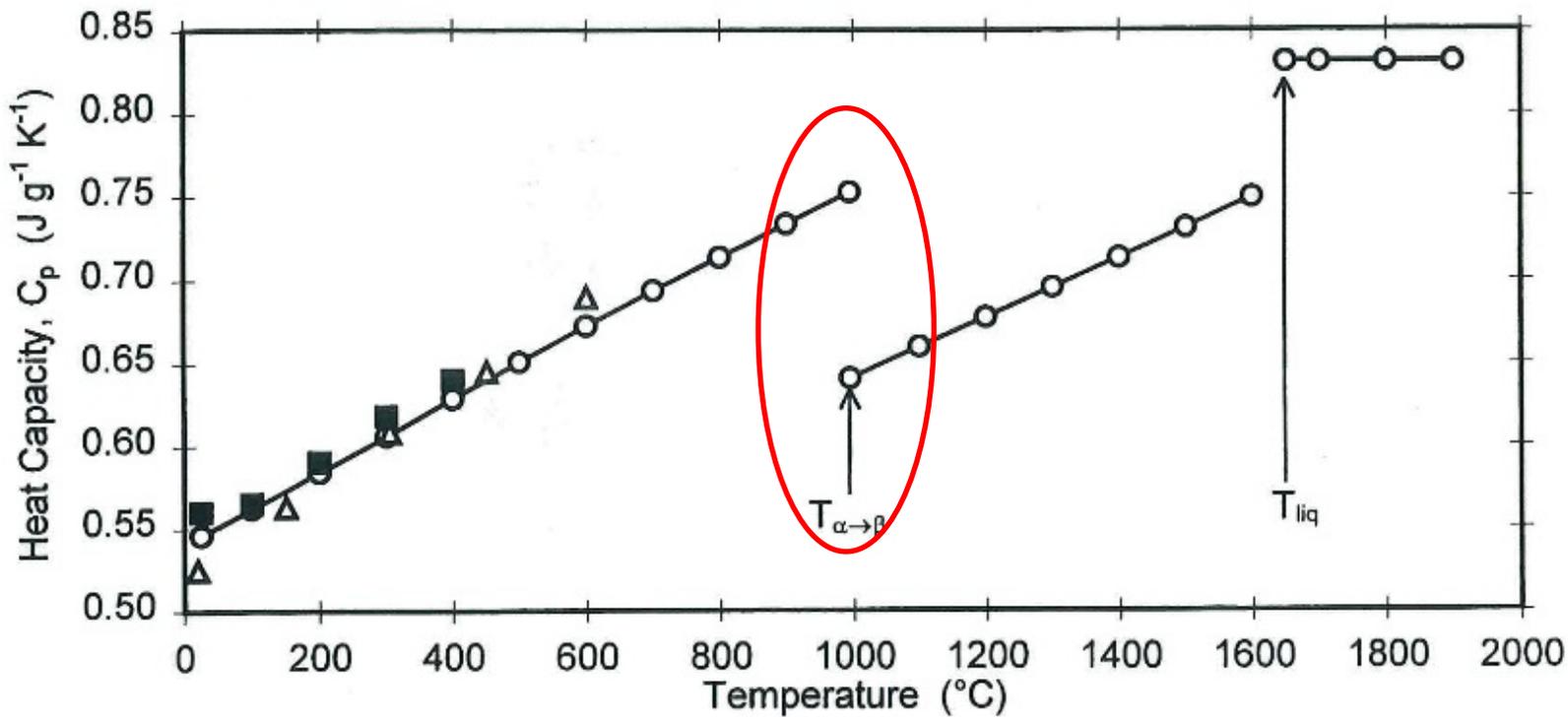
# Models available in OpenFOAM (for a given solver)

(8/8)

heRhoThermo	pureMixture	polynomial	janaf	PengRobinsonGas	specie	sensibleEnthalpy
heRhoThermo	pureMixture	sutherland	hConst	PengRobinsonGas	specie	sensibleEnthalpy
heRhoThermo	pureMixture	sutherland	hConst	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	pureMixture	sutherland	hConst	incompressiblePerfectGas	specie	sensibleInternalEnergy
heRhoThermo	pureMixture	sutherland	hConst	perfectGas	specie	sensibleEnthalpy
heRhoThermo	pureMixture	sutherland	hConst	perfectGas	specie	sensibleInternalEnergy
heRhoThermo	pureMixture	sutherland	janaf	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	pureMixture	sutherland	janaf	incompressiblePerfectGas	specie	sensibleInternalEnergy
heRhoThermo	pureMixture	sutherland	janaf	perfectGas	specie	sensibleEnthalpy
heRhoThermo	pureMixture	sutherland	janaf	perfectGas	specie	sensibleInternalEnergy
heRhoThermo	reactingMixture	const	hConst	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	reactingMixture	const	hConst	incompressiblePerfectGas	specie	sensibleInternalEnergy
heRhoThermo	reactingMixture	const	hConst	perfectGas	specie	sensibleEnthalpy
heRhoThermo	reactingMixture	const	hConst	perfectGas	specie	sensibleInternalEnergy
heRhoThermo	reactingMixture	polynomial	hPolynomial	icoPolynomial	specie	sensibleEnthalpy
heRhoThermo	reactingMixture	polynomial	hPolynomial	icoPolynomial	specie	sensibleInternalEnergy
heRhoThermo	reactingMixture	sutherland	janaf	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	reactingMixture	sutherland	janaf	incompressiblePerfectGas	specie	sensibleInternalEnergy
heRhoThermo	reactingMixture	sutherland	janaf	perfectGas	specie	sensibleEnthalpy
heRhoThermo	reactingMixture	sutherland	janaf	perfectGas	specie	sensibleInternalEnergy
heRhoThermo	singleStepReactingMixture	sutherland	janaf	perfectGas	specie	sensibleEnthalpy
heRhoThermo	singleStepReactingMixture	sutherland	janaf	perfectGas	specie	sensibleInternalEnergy
heRhoThermo	veryInhomogeneousMixture	const	hConst	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	veryInhomogeneousMixture	const	hConst	perfectGas	specie	sensibleEnthalpy
heRhoThermo	veryInhomogeneousMixture	sutherland	janaf	incompressiblePerfectGas	specie	sensibleEnthalpy
heRhoThermo	veryInhomogeneousMixture	sutherland	janaf	perfectGas	specie	sensibleEnthalpy

# Other example of thermophysical model : *phase change in solid state*

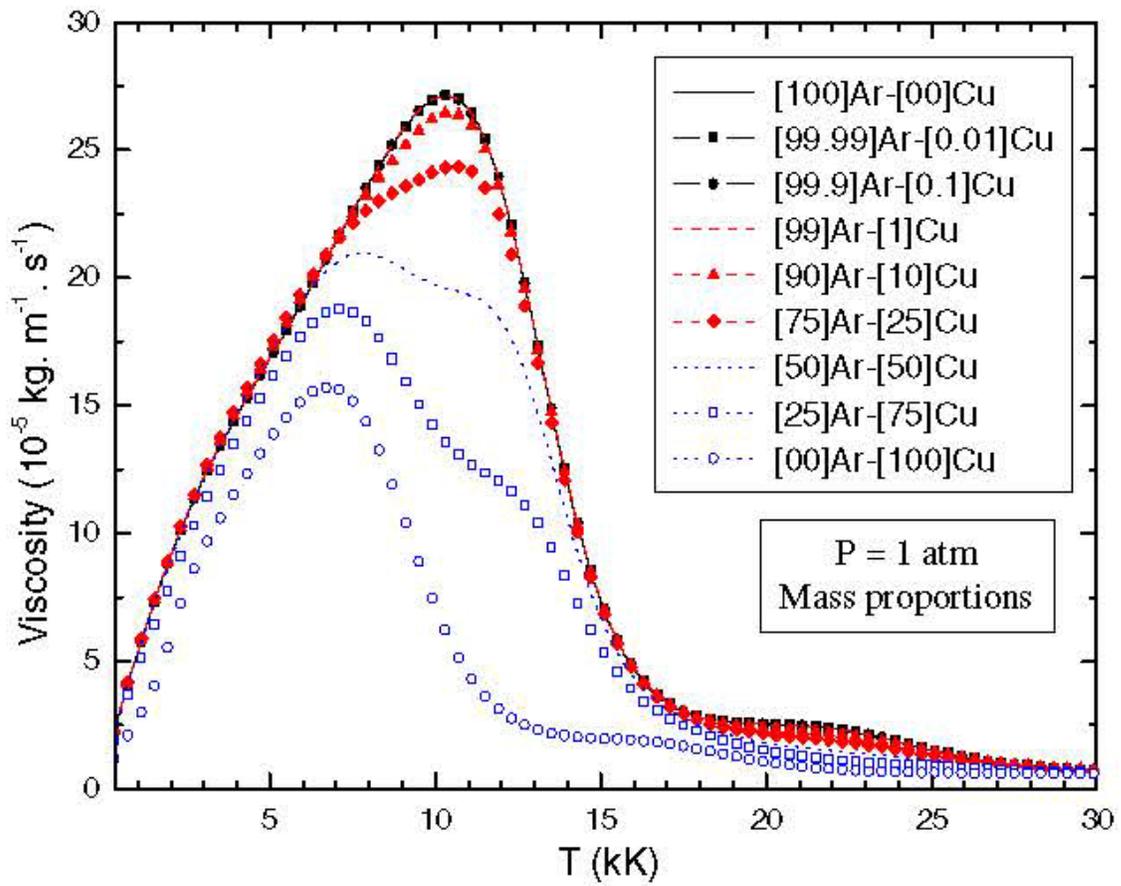
(1/3)



Heat capacity of Ti/6Al/4V as a function of temperature; —, o, recommended values;  $\Delta$ , Bros [3];  $\blacksquare$ , Richardson [4].

From "Recommended values of thermophysical properties for selected commercial alloys", Kenneth C Mills, ASM International, ISBN 0-87170-753-5, (page 213)

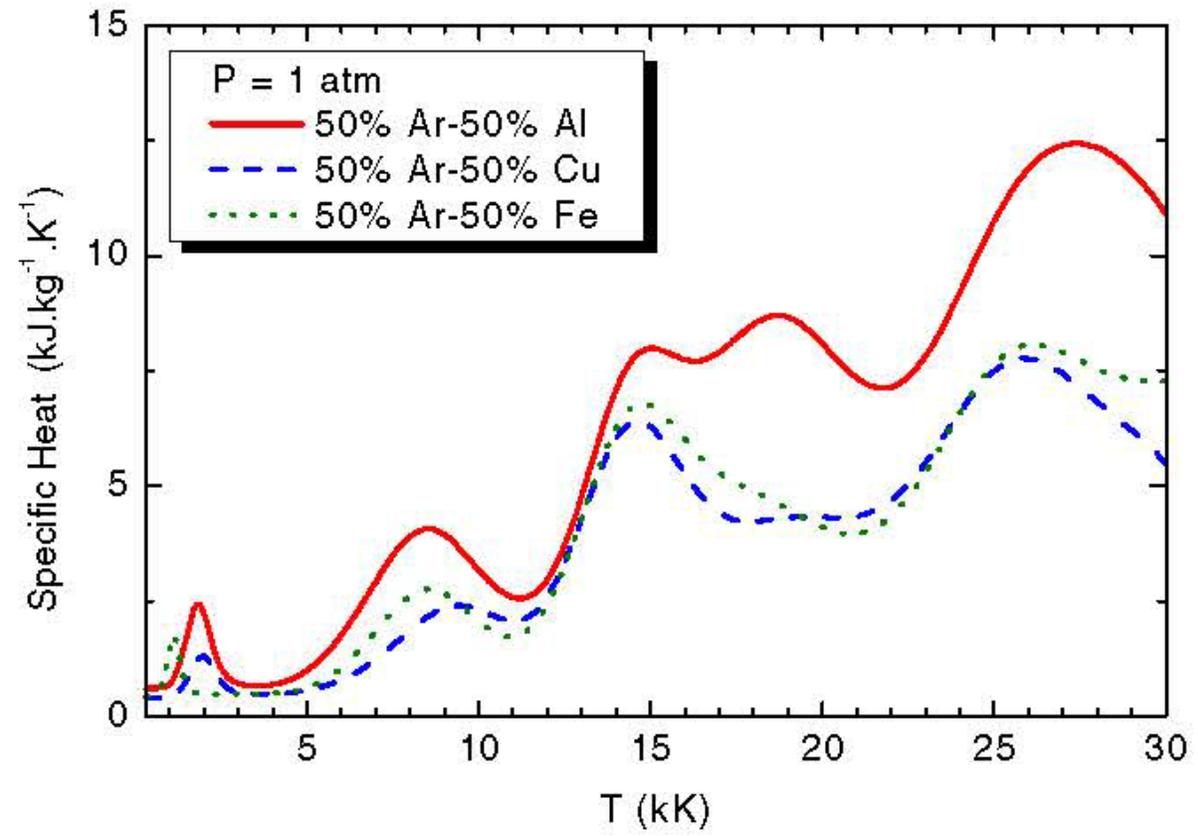
# Other example of thermophysical model : *thermal plasma*



Viscosity of Ar-Cu mixtures at atmospheric pressure.

From “Thermal plasma properties for Ar-Cu, Ar-Fe and Ar-Al mixtures used in welding plasmas processes: II. Transport coefficients at atmospheric pressure”, Y Cressault, A B Murphy, Ph Teulet, A Gleizes and M Schnick, J. Phys. D: Appl. Phys. **46** (2013) 415207

# Other example of thermophysical model : *thermal plasma*



**Figure 1.** Specific heat at constant pressure for 50%Ar–50% metal vapour mixtures by mass at atmospheric pressure.

From “Thermal plasma properties for Ar–Cu, Ar–Fe and Ar–Al mixtures used in welding plasmas processes: II. Transport coefficients at atmospheric pressure”, Y Cressault, A B Murphy, Ph Teulet, A Gleizes and M Schnick, J. Phys. D: Appl. Phys. **46** (2013) 415207

# Example of problem needing a new thermophysical model :

(1/16)

thermal conduction in a high temperature argon gas

Governing equation

$$\frac{\partial(\rho C_v T)}{\partial t} = \nabla(\kappa \nabla \cdot T)$$

with  $\rho$ ,  $C_v$  and  $\kappa$  function of  $T$  obtained

from kinetic theory for  $T \in [200; 20000]K$

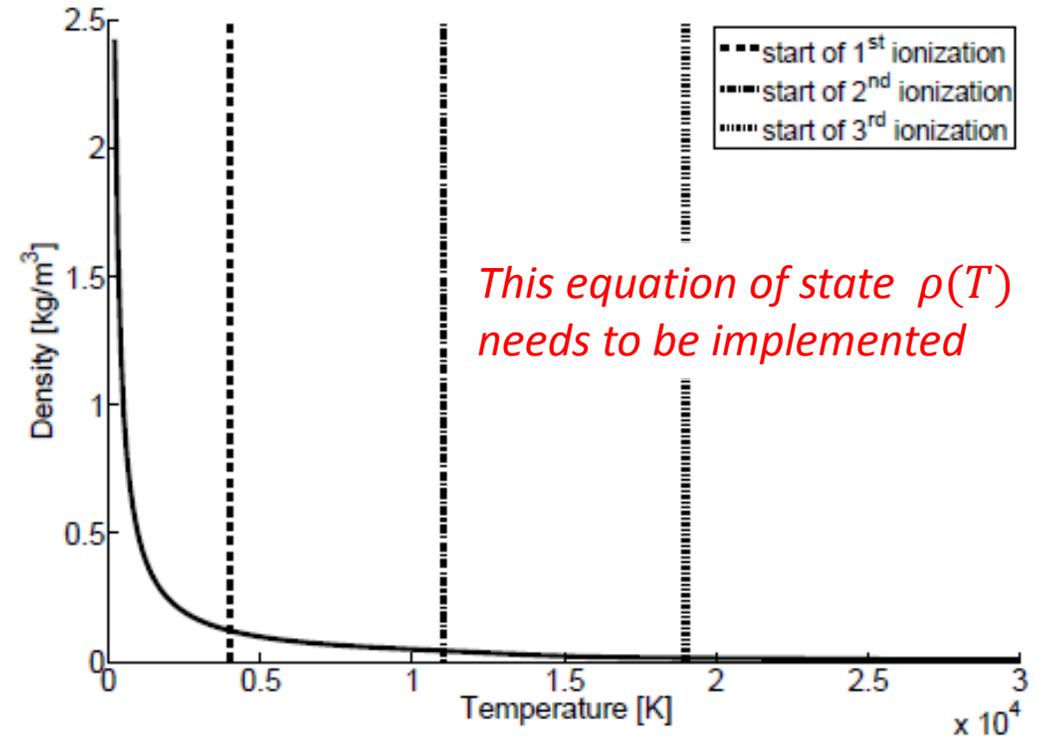


Figure 2.2: Density of an argon gas versus temperature .

*Rk: need a heatTransfer solver*

From "Plasma arc welding simulation with openFOAM",  
M. Sass-Tisovskaya, Lic. thesis, Dept. Applied Mechanics,  
Chalmers University of Technology, 2009 (page 30) 15

# Example of problem needing a new thermophysical model :

(2/16)

thermal conduction in a high temperature argon gas

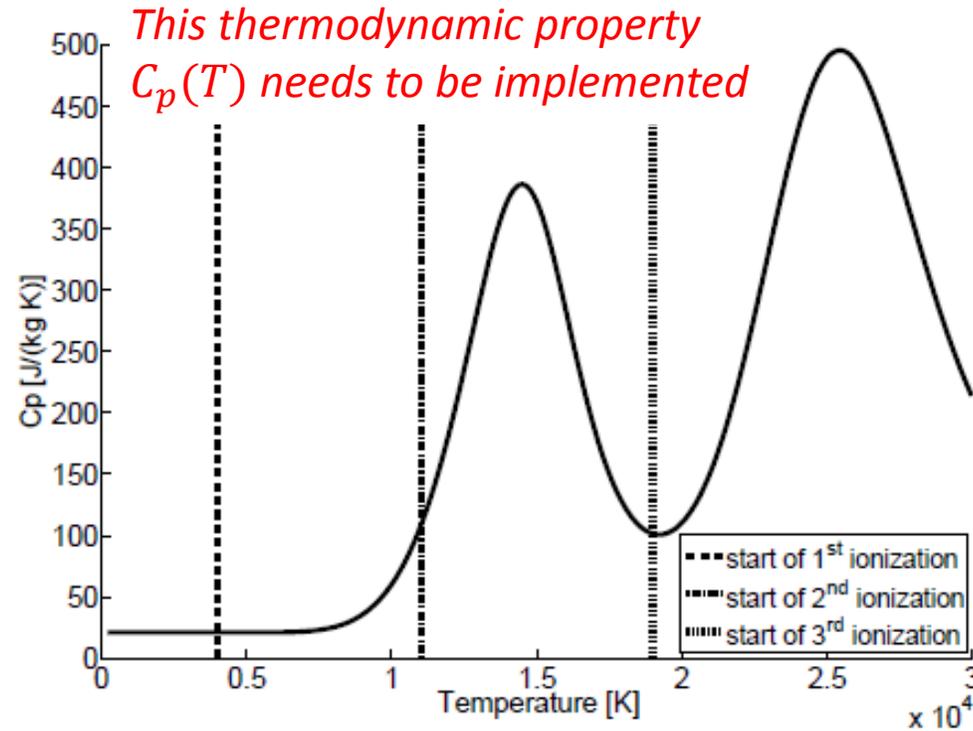


Figure 2.5: Heat capacity at constant pressure of an argon gas versus temperature.

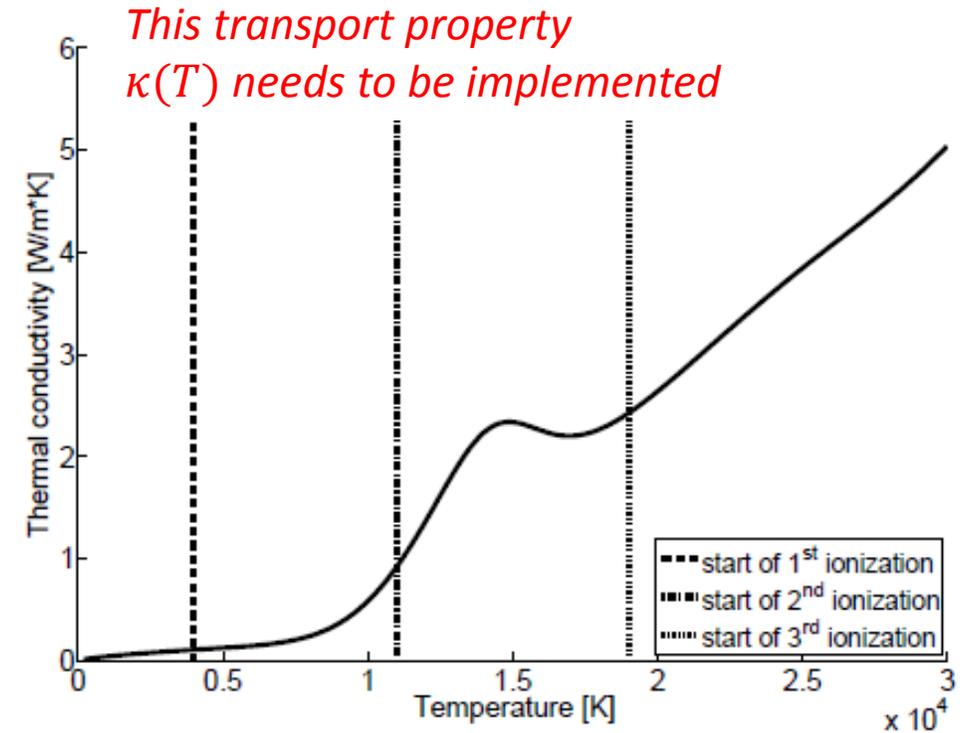


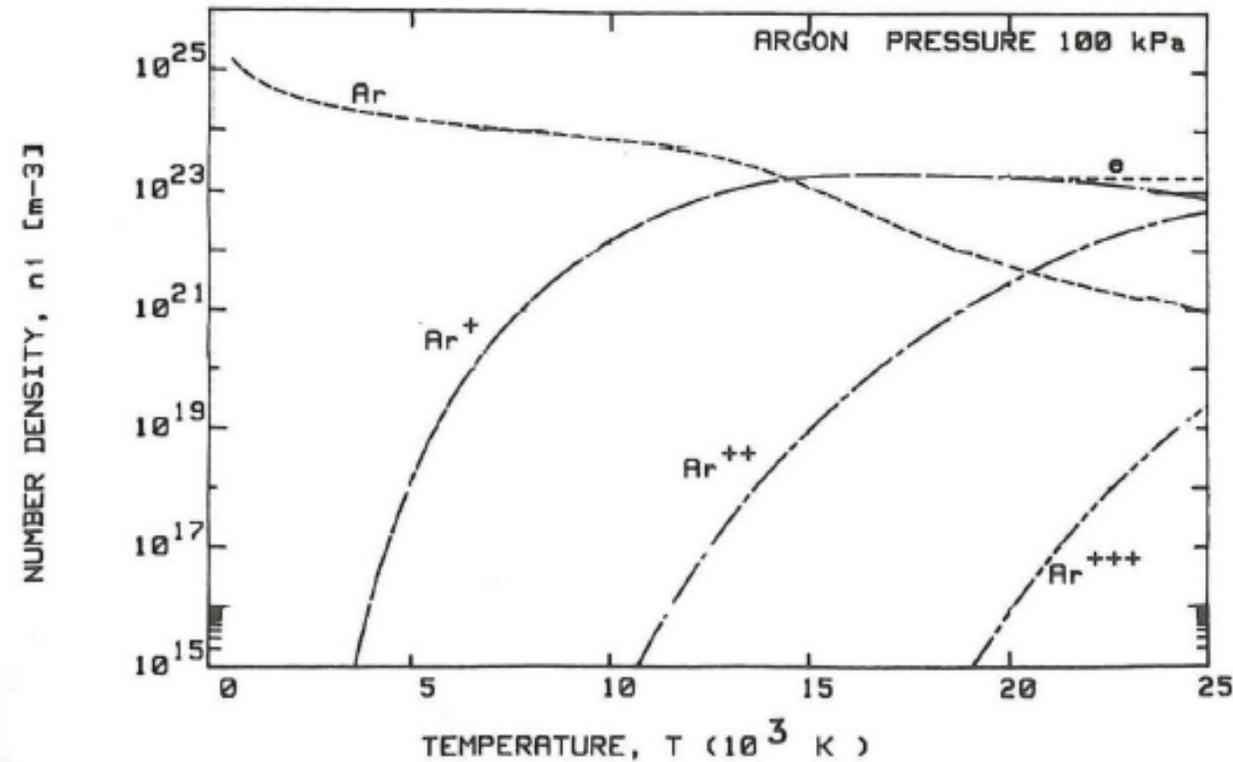
Figure 2.4: Thermal conductivity of an argon gas versus temperature.

*From "Plasma arc welding simulation with openFOAM", M. Sass-Tisovskaya,  
Lic. thesis, Dept. Applied Mechanics, Chalmers University of Technology, 2009 (page 31)*

# Example of problem needing a new thermophysical model :

(3/16)

thermal conduction in a high temperature argon gas



Temperature dependence of the equilibrium composition (species number densities) of an argon plasma at atmospheric pressure (starting from one mole of Ar at room temperature [21].)

*From "Thermal Plasmas, Fundamentals and Applications", Boulos M.I., Fauchais P. and Pfender E.: Vol. 1, Plenum Press, New York, 1994 (page 235)*

Why these peaks in  $C_p$  and  $\kappa$  ?  
Because the fluid is not a pure substance: its composition changes with T.

# Example of problem needing a new thermophysical model :

(4/16)

thermal conduction in a high temperature argon gas

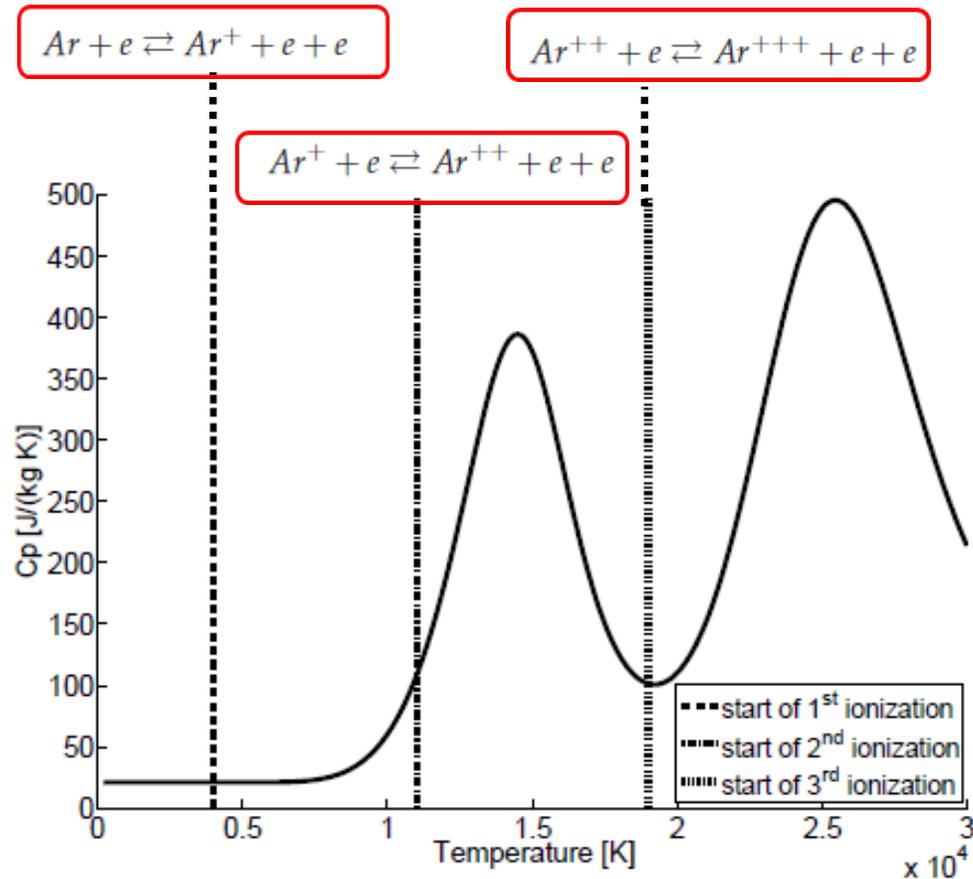


Figure 2.5: Heat capacity at constant pressure of an argon gas versus temperature.

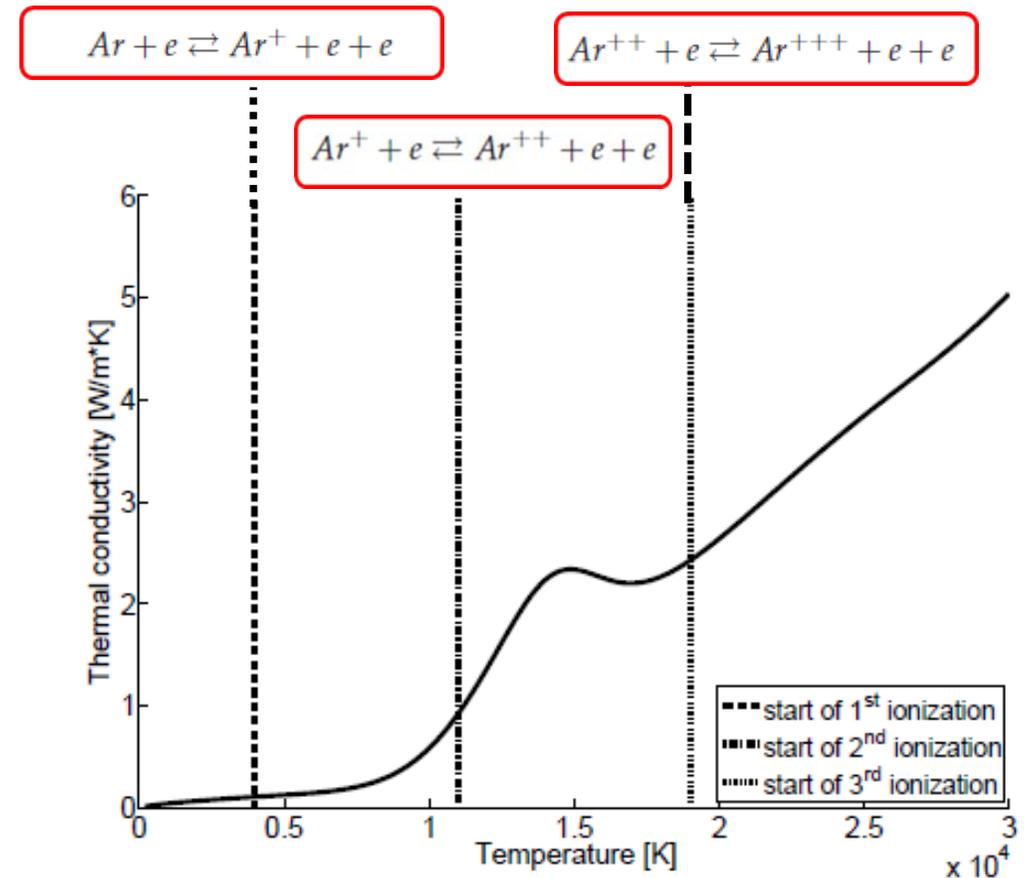


Figure 2.4: Thermal conductivity of an argon gas versus temperature.

From "Plasma arc welding simulation with openFOAM", M. Sass-Tisovskaya,

## Heat transfer and buoyancy-driven flows

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<code>buoyantBoussinesqPimpleFoam</code>	Transient solver for buoyant, turbulent flow of incompressible fluids
<code>buoyantBoussinesqSimpleFoam</code>	Steady-state solver for buoyant, turbulent flow of incompressible fluids
<code>buoyantPimpleFoam</code>	Transient solver for buoyant, turbulent flow of compressible fluids for ventilation and heat-transfer
<code>buoyantSimpleFoam</code>	Steady-state solver for buoyant, turbulent flow of compressible fluids
<code>chtMultiRegionFoam</code>	Combination of <code>heatConductionFoam</code> and <code>buoyantFoam</code> for conjugate heat transfer between a solid region and fluid region
<code>chtMultiRegionSimpleFoam</code>	Steady-state version of <code>chtMultiRegionFoam</code>
<code>thermoFoam</code>	Evolves the thermodynamics on a frozen flow field

- **Copy the solver in your user directory and rename:**

```
cd $WWM_PROJECT_USER_DIR/applications/solvers/heatTransfer
```

*rk: if you do not yet have the heatTransfer directory under /solvers, create it (mkdir heatTransfer)*

```
cp -r $WWM_PROJECT_DIR/applications/solvers/heatTransfer/thermoFoam .
```

```
mv thermoFoam myThermoFoam
```

```
cd myThermoFoam
```

```
mv thermoFoam.C myThermoFoam.C
```

- **Modify Make/files to**

```
myThermoFoam.C
```

```
EXE = $(FOAM_USER_APPBIN)/myThermoFoam
```

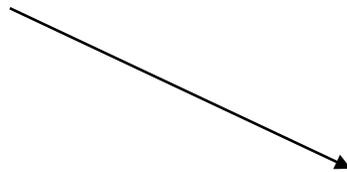
- **Clean and compile**

```
wclean 
```

```
wmake
```

## Parts of thermoFoam involving a thermophysical model

```
createFields.H  
EEqn.H  
Make  
setAlphaEff.H  
thermoFoam.C
```



```
...  
#include "rhoThermo.H"
```

*Include a thermophysical library*



```
...  
#include "EEqn.H"
```

*Include an energy conservation equation*



Part of the energy conservation equation Eeqn.H

```
volScalarField& he = thermo.he();  
  
fvScalarMatrix EEqn  
(  
    fvm::ddt(rho, he) + fvm::div(phi, he)  
    + fvc::ddt(rho, K) + fvc::div(phi, K)  
    + (  
        he.name() == "e"  
        ? fvc::div  
        (  
            fvc::absolute(phi/fvc::interpolate(rho), U),  
            p,  
            "div(phi,v,p)"  
        )  
        : -dpdt  
    )  
    - fvm::laplacian(alphaEff, he)  
    ==  
    radiation->Sh(thermo)  
    + fvOptions(rho, he)  
);
```

→ "he" is either the specific internal energy or the specific enthalpy (choice done when preparing a case, in the dictionary constant/thermophysicalProperties)



**Energy conservation** -total energy of flowing fluid:  $\rho(\hat{h} + K)$  - in a control volume moving at  $\mathbf{v}_b$

$$\frac{d}{dt} \int_V \rho(\hat{h} + K) dV + \int_S \rho(\hat{h} + K) (\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} - \frac{d}{dt} \int_V p dV - \int_S p(\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} = - \int_S \mathbf{q} \cdot d\mathbf{s} - \int_S (p\mathbf{v} + \boldsymbol{\tau} \cdot \mathbf{v}) \cdot d\mathbf{s}$$

Enthalpy :  $\hat{h} = \hat{e} + \frac{p}{\rho}$

Kinetic energy:  $K$

*specific enthalpy  $\hat{h}$  or specific internal energy  $\hat{e}$*

```
fvm::ddt(rho, he) + fvm::div(phi, he)
+ fvc::ddt(rho, K) + fvc::div(phi, K)
+ (
    he.name() == "e"
    ? fvc::div
      (
        fvc::absolute(phi/fvc::interpolate(rho), U),
        p,
        "div(phi,v,p)"
      )
    : -dpdt
  )
- fvm::laplacian(alphaEff, he)
```

**Energy conservation** -total energy of flowing fluid:  $\rho(\hat{h} + K)$  - in a control volume moving at  $\mathbf{v}_b$

$$\frac{d}{dt} \int_V \rho (\hat{h} + K) dV + \int_S \rho (\hat{h} + K) (\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} - \frac{d}{dt} \int_V p dV - \int_S p (\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} = - \int_S \mathbf{q} \cdot d\mathbf{s} - \int_S (p\mathbf{v} + \boldsymbol{\tau} \cdot \mathbf{v}) \cdot d\mathbf{s}$$

*specific kinetic energy K*

```
fvm::ddt(rho, he) + fvc::div(phi, he)
+ fvc::ddt(rho, K) + fvc::div(phi, K)
+ (
    he.name() == "e"
    ? fvc::div
      (
        fvc::absolute(phi/fvc::interpolate(rho), U),
        p,
        "div(phiv,p)"
      )
    : -dpdt
)
- fvm::laplacian(alphaEff, he)
```

**Energy conservation** -total energy of flowing fluid:  $\rho(\hat{h} + K)$  - in a control volume moving at  $\mathbf{v}_b$  (11/16)

$$\frac{d}{dt} \int_V \rho(\hat{h} + K) dV + \int_S \rho(\hat{h} + K)(\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} - \frac{d}{dt} \int_V p dV - \int_S p(\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} = - \int_S \mathbf{q} \cdot d\mathbf{s} - \int_S (p\mathbf{v} + \boldsymbol{\tau} \cdot \mathbf{v}) \cdot d\mathbf{s}$$

```
fvm::ddt(rho, he) + fvm::div(phi, he)
+ fvc::ddt(rho, K) + fvc::div(phi, K)
+ (
```

If "he" is the specific enthalpy " $\hat{h}$ "

```
→ he.name() == "e" This is not true (as he is not "e")
? fvc::div
(
    fvc::absolute(phi/fvc::interpolate(rho), U),
    p,
    "div(phi,v,p)" so this part is not used
)
: -dpdt while this term is calculated
)
- fvm::laplacian(alphaEff, he)
```

**Energy conservation** -total energy of flowing fluid:  $\rho(\hat{h} + K)$  - in a control volume moving at  $\mathbf{v}_b$  (12/16)

$$\frac{d}{dt} \int_V \rho (\hat{h} + K) dV + \int_S \rho (\hat{h} + K) (\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} - \frac{d}{dt} \int_V p dV - \int_S p (\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} = - \int_S \mathbf{q} \cdot d\mathbf{s} - \int_S (p\mathbf{v} + \boldsymbol{\tau} \cdot \mathbf{v}) \cdot d\mathbf{s}$$

$$\hat{h} = \hat{e} + \frac{p}{\rho}$$



If "he" is the specific internal energy  $\hat{e}$

```
fvm::ddt(rho, he) + fvm::div(phi, he)
+ fvc::ddt(rho, K) + fvc::div(phi, K)
+ (
```

$\longrightarrow$  `he.name() == "e"` *This is true*

```
? fvc::div
(
  fvc::absolute(phi/fvc::interpolate(rho), U),
  p,
  "div(phi,v,p)" so this term is calculated
)
```

```
: -dpdt while this part is not used
)
```

```
- fvm::laplacian(alphaEff, he)
```

# Energy conservation -total energy of flowing fluid: $\rho(\hat{h} + K)$ - in a control volume moving at $\mathbf{v}_b$ (13/16)



$$\frac{d}{dt} \int_V \rho (\hat{h} + K) dV + \int_S \rho (\hat{h} + K) (\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} - \frac{d}{dt} \int_V p dV - \int_S p (\mathbf{v} - \mathbf{v}_b) \cdot d\mathbf{s} = - \int_S \mathbf{q} \cdot d\mathbf{s} - \int_S (p\mathbf{v} + \boldsymbol{\tau} \cdot \mathbf{v}) \cdot d\mathbf{s}$$

$$\hat{h} = \hat{e} + \frac{p}{\rho}$$

```
fvm::ddt(rho, he) + fvm::div(phi, he)
+ fvc::ddt(rho, K) + fvc::div(phi, K)
+ (
  he.name() == "e"
  ? fvc::div
  (
    fvc::absolute(phi/fvc::interpolate(rho), U),
    p,
    "div(phiv,p)"
  )
  : -dpdt
)
- fvm::laplacian(alphaEff he)
```

conduction heat flux

**alphaEff** (the thermal diffusivity) is made of 2 contributions:

$$\text{alphaEff} = \mathbf{\text{alpha laminar}} + \text{alpha turbulent}$$

It is set in solver/heatTransfer/thermoFoam/setAlphaEff.H

via the turbulence library : src/turbulenceModels turbulence >alphaEff();



the "turbulenceModels" library is linked to the thermo library src/thermophysicalModels  
 so if the executables are renamed in a new thermophysical library and the library is compiled,  
 they need to be renamed when they appear in the Make/options of the turbulenceModels  
 library and the turbulenceModels library needs also to be compiled (in \$WM\_PROJECT\_USER\_DIR)

*alpha turbulent* is defined in *src/turbulenceModels*

**alpha laminar** is defined in **src/thermophysicalModels** (as transport property)

## What do we need to implement ?

In EEqn.H the heat flux is written :  $q = - \alpha_{\text{Eff}} \nabla h_e$

where the laminar part of  $\alpha_{\text{Eff}}$  is either:

$\kappa / C_p$  if "he" represents the specific enthalpy  $h$  *(we will work with this case and from now assume  $h_e=h$ )*

or

$\kappa / C_v$  if "he" represents the specific internal energy  $e$

So we **need to implement the thermal conductivity**  $\kappa = \kappa(T)$  - plotted slide 16 (right).

the **density, and the specific heat capacity**, respectively plotted slide 15 and 16, so that  $C_p = \rho(T) \cdot c_p(T)$ .

But this is not sufficient since the conservative variable in EEqn.H is the specific enthalpy  $h$  while the thermodynamic and transport properties depend on another variable: the temperature  $T$ .

So we also need to determine  $T$  from  $h$ . This is done solving (with an iterative procedure already implemented in OpenFAOM) the equation of state



$$\Delta h = \int_{T_{ref}}^T c_p(T) dT$$

It implies that we **also need to implement the specific enthalpy**  $h = h(T)$ .

## This implementation can be done in 2 parts

1. First, the implementation of the new transport property  $\kappa$  (the thermal conductivity) for a temperature range from 200K to 20kK. It will be implemented, compiled and tested.

Rk. The resultant thermophysical model will not be consistent from a physical point of view (as it will be associated with constant enthalpy, constant specific heat, ...). But it runs and has the advantage of allowing splitting the implementation work in smaller parts (easier to debug).

2. Implement the new thermodynamic properties  $C_p$  and  $h$  and the new equation of state  $\rho$ . These must be implemented together to be able to derive the temperature  $T$  from  $h(T)$  using  $c_p(T)$ .

Rk. The resultant thermophysical model will then be consistent from a physical point of view. It can be observed that the consistent model runs faster than the non-consistent model implemented in the 1st part. 



**Step 0** : import and rename suited parts of the library "thermophysicalModels"

**Step 1** : declare the new transport property model (see **Ic**)\*

**Step 2** : define the new transport property model (see **Ic**)\*

**Step 3**: declare the new thermophysical model (see **III**)\*\*

**Step 4** : link the new thermophysical model to the solver

**Step 5** : call the new thermophysical model in a test case

\*Slide 6

\*\* Slide 7

*Prepare your library myThermophysicalModels/mySpecie*

- **Copy the folder “specie” of the library in your user directory and rename it:** 

```
cd $WWM_PROJECT_USER_DIR/src/myThermophysicalModels 
```

rk: if you do not yet have the directory myThermophysicalModels under /src, create it with mkdir

```
cp -r $WWM_PROJECT_DIR/src/thermophysicalModels/specie .
```

```
mv specie mySpecie
```

```
cd mySpecie
```



- **Modify the Make/files to:**

```
 mySpecie.C  
LIB = $(FOAM_USER_LIBBIN)/libspezie
```

*Now the executable is in your working space, and your “libspezie” will be accessed in priority (instead of the OpenFOAM executable in \$FOAM\_LIBBIN, even if the name is the same)*

- **Clean & compile**

```
wclean   
wmake libso
```

*Then this name of executable can be kept unchanged. Doing so, no need to import the turbulence library, no need to rename its links (in Make/options) to your thermo library, and no need to recompile the turbulence library (See slide 28): the \$FOAM\_LIBBIN turbulence library will link to your own thermo library.*

*Prepare your library myThermophysicalModels/myBasic*

- **Copy the folder “basic” of the library in your user directory and rename it:**

```
cd $WM_PROJECT_USER_DIR/src/myThermophysicalModels   
cp -r $WM_PROJECT_DIR/src/thermophysicalModels/basic .  
mv basic myBasic  
cd myBasic
```

- **Modify Make/files to:**

myBasic.C 

```
LIB = $(FOAM_USER_LIBBIN)/libfluidThermophysicalModels
```

*Similar to the previous slide*

- Modify the Make/options file to

*(since myBasic needs to be linked to mySpecie at compilation)*

```
EXE_INC = \  
-I$(LIB_SRC)/finiteVolume/lnInclude \  
-I$(WM_PROJECT_USER_DIR)/src/myThermophysicalModels/mySpecie/lnInclude \  
-I$(LIB_SRC)/meshTools/lnInclude
```



(1) Gives the path to access your own files located in mySpecie

```
LIB_LIBS = \  
-L$(FOAM_USER_LIBBIN) \  
-lfiniteVolume \  
-lspecie
```



(2) Indicates that the compiler must 1st look in your own working space (in \$FOAM\_USER\_LIBBIN) to pick the libraries listed below. If not found there (ex. may be lfiniteVolume is not in your space \$FOAM\_USER\_LIBBIN) the compiler will next look in the OpenFOAM space (in \$FOAM\_LIBBIN). If still not found it will complain.



- Clean the dependencies and compile

```
wclean  
wmake libso
```



(3) Can check the path used by reading the messages written on the screen during compilation (see next slide)



# Implement a new transport property K: step 1

(6/12)

Declare the new transport model in user `src/myThermophysicalModels/mySpecie`

- In `mySpecie/include/thermoPhysicsTypes.H` add the following lines

`#include "kineticArTransport.H"` → *To access the new transport model*

```
typedef
kineticArTransport → Name given to the new transport model
<
  species::thermo
  <
    hConstThermo
    <
      perfectGas<specie>
    >,
    sensibleEnthalpy
  >
> kineticArGasHThermoPhysics; → Name given to the new thermophysical model
```

Define the new transport model in user `src/myThermophysicalModels/mySpecie`

- **Copy and rename an existing model:**

*(Prepare the structure)*

```
cd transport
cp -r const kineticAr
cd kineticAr
mv constTransport.C kineticArTransport.C
mv constTransport.H kineticArTransport.H
mv constTransportI.H kineticArTransportI.H
```

open the files one by one and replace

“constTransport” (NOT just “const” !) with “kineticArTransport”

update the “instantiated type name“ in kineticArTransport.H

so look for “instantiated” and below (only there!) replace “return *const*” with “return *kineticAr*”

- **Clean the dependencies (“wclean”) and compile mySpecie (“wmake libso”)**



- **Open kinticArTransportI.H**

```
// Thermal condicivity changed from constant to tabulated data table
```

```
template<class Thermo>
```

```
inline Foam::scalar Foam::kineticArTransport<Thermo>: kappa
```

*thermal conductivity*

```
{
```

```
    const scalar p,
```

```
    const scalar T
```

```
); const
```

```
{
```

```
    // original version:
```

```
    //return this->Cpv(p, T)*mu(p, T)*rPr_;
```

*Comment the original model*

```
    // new version for argon plasma:
```

```
    // Thermal conductivity kappa [W/(m.K)] function of T, for Argon plasma,
```

```
    // tabulated for T from T0=200K to 20000K
```

```
    // with tabulation interval of dT=100K
```

```
    int i_index;
```

```
    scalar dT=100;
```

```
    ...
```

```
    return kappa_T_Argon;
```

```
    // end of kappa version implemented for argon plasma
```

*insert the new model provided in the file Ar\_Data\_ThermalConduct*

```
// Thermal diffusivity for enthalpy [kg/ms]
//
template<class Thermo>
inline Foam::scalar Foam::kineticArTransport<Thermo>::alphah
(
    const scalar p,
    const scalar T
) const
{
    // original version (with Pr constant):
    //return mu(p, T)*rPr_;

    // new version for argon plasma (since Pr is not constant):
    // Pr = mu(p, T)*Cp(p, T)/kappa(p, T)
    // mu(p, T)/Pr = kappa(p, T)/Cp(p, T)
    return kappa(p, T)/this->Cpv(p, T);

    // end of alpha version implemented for argon plasma
}
```

*Comment the original model*

*Write the new model*



- **Clean the dependencies (“wclean”) and compile mySpecie (“wmake libso”)**



*Declare the new thermophysical model in user src/myThermophysicalModels/myBasic*

- **In myBasic/rhoThermo/rhoThermos.C add the following lines**

```
#include "kineticArTransport.H"   
makeThermo  
(  
  rhoThermo,  
  heRhoThermo,  
  pureMixture,  
  kineticArTransport,  
  sensibleEnthalpy,  
  hConstThermo,  
  perfectGas,  
  specie  
);
```

*New combination of Ia, Ib, Ic, II and III (see slides 3 to 7) defining a new thermophysical model*



- **Clean the dependencies (“wclean”) and compile myBasic (“wmake libso”)**



*Call the new thermophysicalModel in a test case*

- Use the test case provided: `blockThermoFoamCase.tgz`
- Run this case with the solver `thermoFoam` (the original one)
- Copy `blockThermoFoam` to `blockNewThermoFoam` and clean
- Update `constant/thermophysicalProperties` to

```
thermoType
{
    type            heRhoThermo;
    mixture         pureMixture;
    transport       kineticAr;    // new model
    //transport     const;
    thermo          hConst;
    equationOfState perfectGas;
    specie          specie;
    energy          sensibleEnthalpy;
}
```

- Run this case with the solver `myThermoFoam` linked to the new thermophysical library
- Compare the results (for instance doing a plot along a suited line you can choose)

## This implementation can be done in 2 parts

1. First, the implementation of the new transport property  $\kappa$  (the thermal conductivity) for a temperature range from 200K to 20kK. It will be implemented, compiled and tested.

Rk. The resultant thermophysical model will not be consistent from a physical point of view (as it will be associated with constant enthalpy, constant specific heat, ...). But it runs and has the advantage of allowing splitting the implementation work in smaller parts (easier to debug).

2. Implement the new thermodynamic properties  $C_p$  and  $h$  and the new equation of state  $\rho$ .

These must be implemented together since the aim is to be able to derive the temperature  $T$  from  $h(T)$  and to calculate the heat capacity  $C_p = \rho(T) \cdot c_p(T)$

Rk. The resultant thermophysical model will then be consistent from a physical point of view. It can also be observed that the consistent model runs faster than the non-consistent model implemented in the 1st step

**Part 1 is done. We now start this 2<sup>nd</sup> part**

**Step 1** : declare (see **Ia, b**) <sup>#,\*</sup> the new thermodynamic properties and the new equation of state

**Step 2** : define (see **Ia**) <sup>#</sup> the new equation of state

**Step 3** : define (see **Ib**)<sup>\*</sup> the new  thermodynamic properties

**Step 4**: declare (see **III**)<sup>\*\*</sup> the new thermophysical model

**Step 5** : link the new thermophysical model to the solver

**Step 6** : call the new thermophysical model in a test case and run

*<sup>#</sup>See slide 4      <sup>\*</sup>See slide 5      <sup>\*\*</sup> See slide 7*

# Implement new thermodynamic properties and equation of state: step 1

(3/16)

☞  
*Declare the new models in user src/myThermophysicalModels/mySpecie*

- ☞ ☞  
▪ **In mySpecie/include/thermoPhysicsTypes.H**  
**add**

```
#include "hKineticArThermo.H"  
#include "rhoKineticAr.H"
```

→ *To access the new thermodynamic model  
& the new equation of state*

**and modify**

```
typedef  
kineticArTransport  
<  
  species::thermo  
  <  
    hConstThermo  
    <  
      perfectGas<specie>  
    >,  
    sensibleEnthalpy  
  >  
> kineticArGasHTermoPhysics;
```

→ *Name given to  
the new  
thermodynamic model  
&  
equation of state*

**to**

```
typedef  
kineticArTransport  
<  
  species::thermo  
  <  
    hKineticArThermo  
    <  
      rhoKineticAr<specie>  
    >,  
    sensibleEnthalpy  
  >  
> kineticArGasHTermoPhysics;
```

*Define a new equation of state in in user src/myThermophysicalModels/mySpecie*

- **Copy and rename an existing model:**

*Prepare the structure*

```
cd equationOfState
```

```
 cp -r perfectGas rhoKineticAr
```

```
cd rhoKineticAr
```

```
mv perfectGas.C rhoKineticAr.C
```

```
mv perfectGas.H rhoKineticAr.H
```

```
 mv perfectGasI.H rhoKinticArI.H 
```

```
open the files one by one and replace 
```

```
“perfectGas” with “rhoKineticAr”
```

```
update the “instantiated type name” in rhoConstAr.H 
```

```
 so look for “instantiated” and below replace “return perfectGas<” “  
with “return rhoKineticAr<” “
```

- Open rhoKineticArI.H and do the following modifications:

```
template<class Specie>
inline Foam::scalar Foam::rhoKineticAr<Specie>::rho(scalar p, scalar T) const
{
    //old model
    //return p/(this->R()*T);

    //new model
    // Density [kg/m^3] function of T, for Argon plasma,
    // tabulated for T from T0=200K to 20000K
    // with tabulation interval of dT=100K

    int i_index;
    scalar dT=100;
    scalar T0=200;
    scalar Temp_Argon;
    scalar rho_T_Argon;

    ...

    return rho_T_Argon;
}
```

*density*

*Comment the original model*

*insert the new model provided in the file density\_Ar\_Data*

- Modify also

```
template<class Specie>
inline Foam::scalar Foam::rhoKineticAr<Specie>::psi(scalar, scalar T) const
{
  // old model
  //return 1.0/(this->R()*T);
  //new model
  Info << " psi should not be used with the rhoKineticAr model " << endl;
  return 0.0;
}
```

*compressibility*

*Comment the original model (ideal gas)*



*Write the new model*  
*Rk: psi is set to zero since the plasma model implemented here is mechanically incompressible, and thermaly expansible:  $\rho(P, T) = \rho(T)$ .*

- **Modify also**

```
template<class Specie>
inline Foam::scalar Foam::rhoKineticAr<Specie>::Z(scalar, scalar) const
{
  // old model
  //return 1.0;

  //new model
  Info << " Z should not be used with the rhoKineticAr model " << endl;
  return 0.0;
}
```

*Compressibility factor*

*Comment the original model (ideal gas)*

*Write the new model*

- **Modify also**

```
template<class Specie>
inline Foam::scalar Foam::rhoKineticAr<Specie>::cpMcv(scalar, scalar) const
{
    // old model
    //return this->RR;

    //new model
    Info << " cpMcv should not be used with the rhoKineticAr model " << endl;
    return 0.0;
}
```

*Comment the original model*

*write the new model*

- **Clean and compile mySpecie**

wclean 

wmake libso

Define the new properties  $h$ ,  $C_p$  in user `src/myThermophysicalModels/mySpecie`

- **Copy and rename an existing model:**

*Prepare the structure*

```
cd thermo
```

```
cp -r hConst hKineticAr
```

```
cd hKineticAr
```

```
mv hConstThermo.C hKineticArThermo.C
```

```
mv hConstThermo.H hKineticArThermo.H
```

```
mv hConstThermoI.H hKineticArThermoI.H
```

open the files one by one and replace

“hConstThermo” with “hKineticArThermo”

update the “instantiated type name” in hKineticArThermo.H

so look for “instantiated” and below replace “return *hConst<*” “  
with “return *hKineticAr<*” “

- Open `hKineticAr.H` and do the following modifications:

```
template<class equationOfState>
inline Foam::scalar Foam::hKineticArThermo<equationOfState>::cp
(
    const scalar p,
    const scalar T
) const
{
```

*Heat capacity  
at constant pressure*

```
// original model
//return Cp_;
```

*Comment the original model*

```
// New model:
// heat capacity at constant pressure [J/(kmol.K)] function of T,
// for Argon plasma,
// tabulated for T from T0=200K to 20000K
// with tabulation interval of dT=100K
```

*insert the new  
model provided in the file  
heatCapacity\_Cp\_Data*

```
int i_index;
scalar dT=100;
...
return Cp_T_Argon*this->W();
```

```
// end of cp version implemented for argon plasma
```

```
}
```

- Modify also

```

template<class equationOfState>
inline Foam::scalar Foam::hKineticArThermo<equationOfState>::ha
(
    const scalar p, const scalar T
) const
{
    // original model
    //return Cp_*T + Hf_;

    // enthalpy [J/kg] function of T, for Argon plasma,
    // tabulated for T from T0=200K to 20000K
    // with tabulation interval of dT=100K

    int i_index;
    scalar dT=100;
    ...
    return h_T_Argon*this->W();
// end of h version implemented for argon plasma
}
    
```

*absolute enthalpy*  
*Hf is the enthalpy of formation*

*Comment the original model*

*insert the new model provided in the file enthalpy\_Data.*  
*Rk. The reference temperature was set so that Hf is zero.*

- **Modify also**

```
template<class equationOfState>
inline Foam::scalar Foam::hKineticArThermo<equationOfState>::hs
(
    const scalar p, const scalar T
) const
{
    // original model
    //return Cp_*T;
    return ha(p,T)-hc();
}
```

*sensible enthalpy*

← *Comment the original constant model*

← *Write the new temperature dependent model.*



*As the non constant thermodynamic models in openFoam depend on both pressure p and temperature, we write ha(p,T) although p is not used.*

- **Modify also**

```
template<class equationOfState>
inline Foam::scalar Foam::hKineticArThermo<equationOfState>::hc() const      chemical enthalpy
{
  // original model
  //return Hf_;
  return 0.;
}
```

*Comment the original model*

*Write the new model*

*Rk. Here the plasma is considered as one-fluid. The ionization reactions were accounted for when tabulating the absolute enthalpy ha. For a one-fluid model hc is the enthalpy of formation. Here it is set to zero (see the remark on slide 53).*



- **Clean (wclean) and compile mySpecie (wmake libso)**

Declare the new thermophysical model in user `src/myThermophysicalModels/myBasic`

- In `myBasic/rhoThermo/rhoThermos.C` add the following lines

```
#include "hKineticArThermo.H"  
#include "rhoKineticAr.H"
```

To access the new thermodynamic model  
& the new equation of state

and change

```
makeThermo  
(  
  rhoThermo,  
  heRhoThermo,  
  pureMixture,  
  kineticArTransport,  
  sensibleEnthalpy,  
  hConstThermo,  
  perfectGas,  
  specie  
);
```

to:

```
makeThermo  
(  
  rhoThermo,  
  heRhoThermo,  
  pureMixture,  
  kineticArTransport,  
  sensibleEnthalpy,  
  hKineticArThermo,  
  rhoKineticAr,  
  specie  
);
```

Name given to  
the new  
thermodynamic model  
&  
equation of state

- Clean (`wclean`) and compile `myBasic` (`wmake libso`)

*Link the new thermophysical model to the solver*



- **The update of myThermoFoam/Make/options** needed to access the new thermophysical library has already been done when implementing the thermal conductivity  $\kappa$  (see slide 41)

So nothing else needs to be done here (unless you decided to rename myBasic in part 2; in that case see slide 41).

*Call the new thermophysical model in a test case and run*

- Copy `blockThermoFoam` to `blockKineticArThermoFoam` and clean (wclean)
- Update constant/thermophysicalProperties to

```
thermoType
{
    type            heRhoThermo;
    mixture         pureMixture;
    transport       kineticAr; // new model
    //transport     const;
    thermo          hKineticAr; // new model
    //thermo        hConst;
    equationOfState rhoKineticAr; // new model
    //equationOfState perfectGas;
    specie          specie;
    energy          sensibleEnthalpy;
}
```

- Run this case with the solver `MyThermoFoam` (now linked to your new thermophysical library)
- Compare the results with the previous results (doing a plot along a suited line you can choose)