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CFD with OpenSource Software  
Assignment 3

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A modified version of the reactingFoam tutorial for LES

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**Author**  
Piero Iudiciani

**Peer reviewers**  
Chen Huang  
Håkan Nilsson

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# 1 Introduction

If one wants to simulate turbulent reacting flows in OpenFOAM, one of the possibilities is to use the the RANS solver *reactingFoam*. However time dependent solutions and resolution of smaller turbulent scales are important in order to capture the flame dynamics. Therefore the main purpose is to transform *reactingFoam* into a LES solver. Files from a LES solver (XiFoam) will be used and changed. All the passages will de described in details. The new solver *reactingFoamLES* will then be applied on a testcase already available for *reactingFoam*. *reactingFoam* solves transport equations for the species involved in the combustion. In the tutorial the files which set the reaction mechanism will be described, and it will be shown how to select a different one. This tutorial works on OpenFoam version 1.5.x, thus we have to source it:

```
source $FOAM_INST_DIR/OpenFOAM-1.5.x/etc/bashrc
```

## 2 Creating a new solver

We copy the source of the available solver *reactingFoam* from the \$FOAM\_APP folder:

```
cd $FOAM_USER_APPBIN
cp -r $FOAM_APP/solvers/combustion/reactingFoam reactingFoamLES
cd reactingFoamLES
```

Then we copy some files from the course homepage using an internet browser or *wget*:

```
wget http://www.tfd.chalmers.se/~hani/kurser/OS_CFD_2009/ \
PieroIudiciani/tutFiles.tar.gz
tar xzf tutFiles.tar.gz
rm tutFiles.tar.gz
```

First we rename the source file *reactingFoam.C* into *reactingFoamLES.C* and we edit it:

```
mv reactingFoam.C reactingFoamLES.C
```

In particular we change line 35 from

```
#include "compressible/RASModel/RASModel.H"
```

to

```
#include "compressible/LESModel/LESModel.H"
```

and we add the following at line 40: (all the LES solvers, i.e. *coodles*, *XiFoam* have this line in the source file)

```
#define divDevRhoReff divDevRhoBeff
```

In `createFields.H` we substitute RAS with LES in lines 61-63. We also comment lines 72-74. According to a low Mach number approximation, in fact, in the following we will neglect variations in time of the pressure in the discretization of the equations:

---

```
Info << "Creating turbulence model." << nl;
autoPtr<compressible::RASModel> turbulence
(
    compressible::RASModel::New
    (
        rho,
        U,
        phi,
        thermo()
    )
);

Info<< "Creating field DpDt" << endl;
volScalarField DpDt =
    fvc::DDt(surfaceScalarField("phiU", phi/fvc::interpolate(rho)), p);}
```

---

becomes:

---

```
Info << "Creating turbulence model" << nl;
autoPtr<compressible::LESModel> turbulence
(
    compressible::LESModel::New
    (
        rho,
        U,
        phi,
        thermo()
    )
);
```

```
//Info<< "Creating field DpDt\n" << endl;
//volScalarField DpDt =
//    fvc::DDt(surfaceScalarField("phiU", phi/fvc::interpolate(rho)), p);
```

---

We now need files in which the discretization of the Navier-Stokes equations is performed. We can copy the pressure, momentum and energy equations from the solver *XiFoam*:

```
cp $FOAM_APP/solvers/combustion/XiFoam/[phU]Eqn.H .
```

and slightly modify them. Basically we comment out the terms containing the variation in time of the pressure. In the pressure equation *p.Eqn.H* we comment out the last line (line68):

```
//DpDt = fvc::DDt(surfaceScalarField("phiU", phi/fvc::interpolate(rho)), p);
```

Similarly in the energy equation *hEqn.H* we comment out line 8 and move one term to the right hand side:

---

```
{
    solve
    (
        fvm::ddt(rho, h)
        + mvConvection->fvmDiv(phi, h)
        - fvm::laplacian(turbulence->alphaEff(), h)
        ==
        DpDt
    );

    thermo->correct();
}
```

---

becomes:

---

```
{
    solve
    (
        fvm::ddt(rho, h)
        + mvConvection->fvmDiv(phi, h)
//    - fvm::laplacian(turbulence->alphaEff(), h)
```

```

    ==
//      DpDt
      fvm::laplacian(turbulence->alphaEff(), h)
    );

    thermo->correct();
}

```

---

In the momentum equation we can neglect the gravitational forces and thus we delete lines 6-7:

---

```

    fvVectorMatrix UEqn
    (
        fvm::ddt(rho, U)
      + fvm::div(phi, U)
      + turbulence->divDevRhoReff(U)
    ==
        rho*g
    );

    if (momentumPredictor)
    {
        solve(UEqn == -fvc::grad(p));
    }

```

---

becomes:

---

```

    fvVectorMatrix UEqn
    (
        fvm::ddt(rho, U)
      + fvm::div(phi, U)
      + turbulence->divDevRhoReff(U)
    );

    if (momentumPredictor)

```

```
{
    solve(UEqn == -fvc::grad(p));
}
```

---

These equation files are also found in the tutorial files:

```
ls tutFiles/*Eqn.H .
```

Then we change the file `Make/files` so that:

```
reactingFoamLES.C
```

```
EXE = $(FOAM_USER_APPBIN)/reactingFoamLES
```

In the file `Make/options` again we change the RAS libraries with the LES ones. We can add the following at line 5:

```
-I$(LIB_SRC)/turbulenceModels/LES/LESdeltas/lnInclude\  
and delete line 2: -I../XiFoam\  

```

---

```
EXE_INC = \  
-I../XiFoam \  
-I$(LIB_SRC)/finiteVolume/lnInclude \  
-I$(LIB_SRC)/turbulenceModels/RAS \  
-I$(LIB_SRC)/thermophysicalModels/specie/lnInclude \  

```

```
[...]
```

```
EXE_LIBS = \  
-lcompressibleRASModels \  
-lcombustionThermophysicalModels \  

```

---

becomes:

---

```
EXE_INC = \  
-I$(LIB_SRC)/finiteVolume/lnInclude \  
-I$(LIB_SRC)/turbulenceModels/LES \  
-I$(LIB_SRC)/turbulenceModels/LES/LESdeltas/lnInclude \  

```

[...]

```
EXE_LIBS = \  
  -lcompressibleLESModels \  
  -lcombustionThermophysicalModels \  
  -lOpenFOAM
```

---

We are now ready to compile:

```
wclean  
rm -r Make/linuxGccDP*  
wmake
```

### 3 Setting up a case

Once we have the solver, we can set up a case. A tutorial file for *reactingFoam* is found in the OpenFOAM wiki. Either we download from the webpage [http://openfoamwiki.net/index.php/Tut\\_reactingFoam\\_firstTutorial](http://openfoamwiki.net/index.php/Tut_reactingFoam_firstTutorial) or we get it by typing directly in the shell:

```
wget http://openfoamwiki.net/images/b/b6/ReactingFoamCase.tar.gz
```

Once downloaded we extract the files in the folder *ReactingFoamCase*:

```
run  
mkdir ReactingFoamCase  
mv ReactingFoamCase.tar.gz ReactingFoamCase  
cd ReactingFoamCase  
tar xzf ReactingFoamCase.tar.gz  
rm ReactingFoamCase.tar.gz
```

#### 3.1 The *chemkin* folder

We can see that we have a folder called *chemkin*. This folder contains the informations for the chemical reaction mechanism.

```
ls chemkin/  
chem.inp  chem.inp.1  chem.inp_15  chem.inp.full  therm.dat
```

##### **chem.inp file**

When solving flows with reactions, a transport equation for each of the species involved is also solved. The files *\*inp\** contain the information about the species and the reaction mechanism. A reaction mechanism is a list of reactions that occur during a combustion



process. Each reaction is characterized by a reaction rate  $k_i$  which is the "speed" at which the reaction occurs and is characterized by the Arrhenius equation:

$$k_i = A_i T^{\beta_i} \exp\left(-\frac{E_i}{RT}\right) \quad (1)$$

where  $E_i$  is the energy of activation,  $A_i$  and  $\beta_i$  are experimental parameters,  $T$  is the temperature. A detailed mechanism containing all the species is composed by hundreds of species and reactions. It is therefore not practically solvable and usually reduced mechanism with very few reactions are used. The simplest mechanism possible is composed by only one global reaction. This is the case of the *chem.inp* file which contains a simple one-step reaction mechanism for heptane (C7H16):

```

ELEMENTS
  H   O   C N
END
SPECIE
C7H16 O2 N2 CO2 H2O
END
REACTIONS
  C7H16 + 11O2          => 7CO2 + 8H2O          5.00E+8  0.0  15780.0!  1
      FORD      / C7H16 0.25 /
      FORD      / O2 1.5 /
END

```

In this mechanism the chemical elements involved are hydrogen (H), oxygen (O), carbon (C) and nitrogen (N). The species that participate are heptane (C7H16), oxygen molecule (O2), nitrogen molecule (N2), carbon dioxide (CO2), water, (H2O). In this case there is only one reaction in which heptane is the fuel, oxygen is the oxidizer, CO2 and water are the products. Nitrogen is the inert species. The three numbers before the question mark carry the information about the reaction rate and represent respectively the parameters  $A_i$ ,  $\beta_i$  and  $dE_i$  in 1. An example of a more complex mechanism can be found in *chem.inp.full*

### therm.dat file

The file *therm.dat* instead contains a database of coefficients for several species. Such coefficients are needed to compute thermodynamical variables such as specific heat  $c_p/R$ , enthalpy  $H^0/RT$ , entropy  $S^0/R$ , according to the following equations:

$$c_p/R = a_1 + a_2 T + a_3 T^2 + a_4 T^3 + a_5 T^4 \quad (2)$$

$$H^0/RT = a_1 + \frac{a_2}{2} T + \frac{a_3}{3} T^2 + \frac{a_4}{4} T^3 + \frac{a_5}{5} T^4 + \frac{a_6}{T} \quad (3)$$

$$S^0/R = a_1 \ln T + \frac{a_2}{2} T + \frac{a_3}{2} T^2 + \frac{a_4}{3} T^3 + \frac{a_5}{4} T^4 + a_7 \quad (4)$$

The *therm.dat* file looks like this:

```

THERMO ALL
  200.000  1000.000  5000.000
(CH2O)3          70590C   3H   60   3   G  0200.00  4000.00  1500.00    1
  0.01913678E+03 0.08578044E-01-0.08882060E-05-0.03574819E-08 0.06605143E-12    2
-0.06560876E+06-0.08432507E+03-0.04662286E+02 0.06091547E+00-0.04710536E-03    3
  0.01968843E-06-0.03563271E-10-0.05665404E+06 0.04525265E+03    4
(CH3)2SICH2      61991H   8C   3SI  1   G  0200.00  2500.00  1500.00    1
  0.01547852E+03 0.01065700E+00-0.01234345E-05-0.01293352E-07 0.02528715E-11    2
-0.06693076E+04-0.05358884E+03 0.02027522E+02 0.04408673E+00-0.03370024E-03    3
  0.01484466E-06-0.02830898E-10 0.03931454E+05 0.01815821E+03    4
AL                62987AL  1           G  0200.00  5000.00  0600.00    1
  0.02559589E+02-0.01063224E-02 0.07202828E-06-0.02121105E-09 0.02289429E-13    2
  0.03890214E+06 0.05234522E+02 0.02736825E+02-0.05912374E-02-0.04033938E-05    3
  0.02322343E-07-0.01705599E-10 0.03886795E+06 0.04363880E+02    4
[...]
```

These files are written and organized according to chemkin software format. Details can be found in<sup>1</sup> and are here resumed. The first line is chemkin syntax necessary at the beginning of the file. The three values in the second line specify three values of temperature and therefore two temperature ranges. The intermediate temperature is generally always 1000K. For each species then four lines are reported. The first species is in this case (CH<sub>2</sub>O)<sub>3</sub> and does not necessarily need to be used in the reaction mechanism. The entries in the first line report respectively the name of the species, its elemental composition, its electronic composition, its phase (G for gas, L for liquid, S for solid), and three temperatures (low, high, intermediate). The fourteen entries in the following 3 lines report the 7 coefficients  $a_{1-7}$  in equations 2-4 for the two temperature ranges, (higher range and lower range respectively).

The chemkin directory should be located in *constant*:

```
mv chemkin/ constant/
```

## 3.2 The *constant* folder

In the folder *constant* the following files are found:

```
ls constant/
chemistryProperties  environmentalProperties  turbulenceProperties
chemkin              polyMesh
combustionProperties  thermophysicalProperties
```

<sup>1</sup>[http://www.tfd.chalmers.se/~hani/kurser/OS\\_CFD\\_2007/AndreasLundstrom/reactingFoam.pdf](http://www.tfd.chalmers.se/~hani/kurser/OS_CFD_2007/AndreasLundstrom/reactingFoam.pdf)  
[http://www.tfd.chalmers.se/~hani/kurser/OS\\_CFD\\_2008/PerCarlsson/PC\\_Tutorial\\_dieselFoam\\_peered\\_NL\\_HN.pdf](http://www.tfd.chalmers.se/~hani/kurser/OS_CFD_2008/PerCarlsson/PC_Tutorial_dieselFoam_peered_NL_HN.pdf)

## thermophysicalProperties file

In the constant/thermophysicalProperties file we change the path of the chemical files:

---

```
/* * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * */  
  
thermoType hMixtureThermo<reactingMixture>;  
  
CHEMKINFile           "/cluster/samples/reactingFoam/chemkin/chem.inp";  
CHEMKINThermoFile     "/cluster/samples/reactingFoam/chemkin/therm.dat";
```

---

becomes:

---

```
/* * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * */  
  
thermoType hMixtureThermo<reactingMixture>;  
  
CHEMKINFile           "chemkin/chem.inp";  
CHEMKINThermoFile     "chemkin/therm.dat";
```

---

## chemistryProperties file

The settings for the *chemistryProperties* file are as follows:

```
// * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * //  
  
chemistry             on;  
  
turbulentReaction off;  
Cmix                  Cmix [ 0 0 0 0 0 0 0 ] 1.0 ;  
  
//chemistrySolver     ODE;  
//chemistrySolver     EulerImplicit;  
chemistrySolver       sequential;  
  
initialChemicalTimeStep 1.0e-8;
```



```

// * * * * *
application      reactingFoamLES;

startFrom        latestTime;

startTime        0;

stopAt           endTime;

endTime          1;

deltaT           1e-04;

writeControl     adjustableRunTime;

writeInterval    1.0e-3;

purgeWrite       0;

writeFormat      binary;

writePrecision   6;

writeCompression uncompressed;

timeFormat       general;

timePrecision    6;

adjustTimeStep   yes;

maxCo            0.1;

runTimeModifiable yes;

// *****

```

### 3.4 The *0/* directory and boundary conditions

We remove the latest directory, and in the *0/* directory we create a file for the turbulent viscosity *muSgs* which is needed by the Smagorinsky model. The right dimensions should be assigned to the turbulent viscosity:

```

rm -r 0.055/
cp 0/k 0/muSgs

// * * * * *

dimensions      [ 1 -1 -1 0 0 0 ];

internalField   uniform 0.0;

boundaryField
{
    inlet
    {
        type          fixedValue;
        value uniform 0.0;
    }
    lowerInlet
    {
        type          fixedValue;
        value uniform 0.0;
    }

    outlet
    {
        type          zeroGradient;
    }

    upperWall
    {
        type          zeroGradient;
    }
}

```

Make sure that *blockMesh* has been run and then run *reactingFoamLES*

```
reactingFoamLES > log &
```

## 4 Use another Reaction Mechanism

One might want to simulate different fuels or use a more detailed mechanism. In the tut-Files directory for example the Westbrook and Dryer two-steps mechanism for methane is available:

```
cp $FOAM_USER_APPBIN/reactingFoamLES/tutFiles/WD2steps.inp constant/chemkin/
```

```
ELEMENTS
C H O N
END
SPECIE
CH4 O2 CO H2O CO2 N2
END
REACTIONS
  CH4 + 1.5O2 => CO + 2H2O          2.80E+09      0.0      48400.
    FORD /CH4 -0.3/
    FORD /O2 1.3/
  CO2 => CO + 0.5O2                5.00E+08      0.0      40000.
    FORD /CO2 1.0 /
  CO + H2O + 0.5O2 => CO2 + H2O    3.98E+14      0.0      40000.
    FORD /CO 1.0 /
    FORD /H2O 0.5 /
    FORD /O2 0.25/
END
```

One can see that in this mechanism carbon monoxide (CO) is first formed as intermediate species and then carbon dioxide. Therefore in the 0/ directory we should change the fuel from heptane (C7H16) to methane(CH4) and additionally create files for carbon monoxide, carbon dioxide and water (optionally since any other species is treated by the Ydefault file).

```
cp 0/C7H16 0/CH4
```

Accordingly the *constant/thermophysicalProperties* file should be updated:

```
/* * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * * */

thermoType hMixtureThermo<reactingMixture>;

//CHEMKINFile          "chemkin/chem.inp";
CHEMKINFile            "chemkin/WD2steps.inp";
CHEMKINThermoFile     "chemkin/therm.dat";

inertSpecie            N2;
```

It is now possible to run with the new mechanism.

```
rm 0.*/  
reactingFoamLES > log2 &
```